

11

CHEMICAL REACTIONS

Practice Problems

In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
4. Write a balanced equation for the production of HCl gas from its elements.
5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



7. Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
8. Write a balanced equation from this word equation.
aqueous silver nitrate + copper metal $\rightarrow$ silver metal + aqueous copper nitrate
9. Write a balanced equation for the following word equation.



SECTION 11.2 TYPES OF CHEMICAL REACTIONS

1. Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
2. Write the balanced equation for the reaction that occurs between aluminum and fluorine.
3. Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
4. Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
5. Write the balanced equation for the combustion of propane (C_3H_8) to produce carbon dioxide and water vapor.
6. Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets

L Reisser



Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets:

The book delves into Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets. Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets
 - Chapter 2: Essential Elements of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets
 - Chapter 3: Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets in Everyday Life
 - Chapter 4: Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets. The first chapter will explore what Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets is, why Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets is vital, and how to effectively learn about Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets.
 3. In chapter 2, the author will delve into the foundational concepts of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets. This chapter will elucidate the essential principles that must be understood to grasp Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets in its entirety.
 4. In chapter 3, this book will examine the practical applications of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets in daily life. This chapter will showcase real-world examples of how Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets in specific contexts. This chapter will explore how Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Prentice Hall Chemistry Chapter 11 Chemical Reactions Worksheets.

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