

Name _____ Class _____ Date _____

Section 2–3 Carbon Compounds (pages 44–48)



Key Concept

- What are the functions of each group of organic compounds?

The Chemistry of Carbon (page 44)

- How many valence electrons does each carbon atom have? Each carbon atom has four electrons.
- What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms.

Macromolecules (page 45)

- Many of the molecules in living cells are so large that they are known as macromolecules.
- What is the process called by which macromolecules are formed? Polymerization
- When monomers join together, what do they form? Polymers
- What are four groups of organic compounds found in living things?
 - Carbohydrates
 - Lipids
 - Nucleic acids
 - Proteins

Carbohydrates (pages 45–46)

- What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbohydrates.
- Circle the letter of each sentence that is true about carbohydrates.
 - ☒ Starches and sugars are examples of carbohydrates.
 - ☒ Living things use them as their main source of energy.
 - The monomers in sugar polymers are starch molecules.
 - ☒ Plants and some animals use them for strength and rigidity.
- Single sugar molecules are also called monosaccharides.
- Circle the letter of each monosaccharide.

☒ galactose	☒ glucose
b. glycogen	☒ fructose

Prentice Hall Chemistry Ch 1 test Answers

**Peter Atkins,Julio de Paula,Ronald
Friedman**



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