

Section 1-2 How Scientists Work (pages 8-14)

Answer each of the following questions:

- In your head first
- Look back into the book/your notes and check to see if what you are thinking is correct
- Write your answer on the paper

Designing an Experiment (pages 8-10)

1. The idea that life can arise from nonliving matter is called _____.
2. What was Francesco Redi's hypothesis about the appearance of maggots?

3. What are variables in an experiment? _____
4. Ideally, how many variables should an experiment test at a time? _____
5. What is a controlled experiment? _____

6. The illustration below shows the beginning of Redi's experiment. Complete the illustration by showing the outcome.



Prentice Hall Biology Work Answers Chapter 4

**Thirumalaisamy P.
Velavancorresponding**



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise

calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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