

Preliminary Chemistry Notes - 2020

Glossary of key terms:

Decomposition - The process by which organic substances are broken down into simpler organic matter

Lustre – How shiny a substance appears and relates to how well the substance is able to reflect light; in general, metals are lustrous, and non-metals are dull

Matter – The material substance that makes up the physical universe

Molecule - A group of two or more atoms covalently bonded together (usually refers to non-metals)

- Molecules can also be defined as particles which move independently of each other
- Thus, noble gases (group 8) which are chemically stable usually exists as single particles moving independently aka monoatomic molecules (containing one atom)
- Diatomic molecules are molecules with two atoms, e.g. H_2 , Cl_2

Compound - Molecules that have at least two elements different, chemically combined in a specific ratio. They are represented by a chemical formula

Mixture – A substance of variable composition, containing two or more elements or compounds that are not chemically bonded together

Covalent Bond – A covalent bond, also called a molecular bond, is a chemical bond that involves the sharing of electron pairs between atoms

Period - A period in the periodic table is a row of chemical elements. All elements in a row have the same number of electron shells. Each next element in a period has one more proton and is less metallic than its predecessor

Group - The periodic table also has a special name for its vertical columns. Each column is called a group. The elements in each group have the same number of electrons in the outer orbital. Those outer electrons are also called valence electrons

Ion – Atom or molecule with a net electrical charge

Polyatomic Ion - A polyatomic ion, also known as a molecular ion, is a **charged** chemical species composed of two or more atoms covalently bonded

Solution – A homogeneous mixture of two or more substances, with a minor substance being uniformly distributed in a major substance. The minor substance is called the solute and the major substance is called the solvent. An example is saltwater (where salt = solute, and water = solvent). Also, the solute is dissolved in the solvent, and solute particles cannot be seen under a light microscope

Preliminary Chemistry Notes

Sabine Zange



Preliminary Chemistry Notes:

Whispering the Strategies of Language: An Emotional Quest through **Preliminary Chemistry Notes**

In a digitally-driven world wherever monitors reign great and instant communication drowns out the subtleties of language, the profound strategies and psychological subtleties concealed within words frequently go unheard. However, situated within the pages of **Preliminary Chemistry Notes** a captivating literary prize sporting with raw feelings, lies an extraordinary quest waiting to be undertaken. Written by an experienced wordsmith, this wonderful opus encourages viewers on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the cloth of every word. Within the emotional depths of this emotional review, we will embark upon a genuine exploration of the book's core styles, dissect its captivating publishing design, and fail to the strong resonance it evokes strong within the recesses of readers' hearts.

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