Predicting Reaction Products

Predict the products for the following reactions, balance the equation, then classify the type of reactions:

- ___Na+__FeBr₃→ 1)
- ___ NaOH + ___ H₂SO4 → 2)
- 3) C₂H₄O₂ + O₂ →
- NH₃ + H₂O → 40
- 5) PbSO. + AgNO₃ →
- 60 PBr_b →
- 7) HBr+ Fe→
- 8) $_{\text{L}}$ KMnO₄ + $_{\text{L}}$ ZnCl₂ \rightarrow
- 99 $MnO_2 + Sn(OH)_4 \rightarrow$
- O₂ + C₂H₂₂O₂ → 10)
- 11) H₂O₂ →
- 12) PtCl₄ + Cl₂ →

Predicting Reaction Products Worksheet Answers

Society of American Value Engineers

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