Answer Key

Precipitation Lab

Purpose:

- 1. Identify the ions present in various aqueous solutions.
- Combine solutions and identify the reactions that form precipitates.
- Write balanced molecular equation, complete ionic equation, and a net ionic equation for each precipitation reaction.



Materials:

- Small test tubes
- Various solutions

Procedure:

- 1. Add 10-15 drops of the each reactant to a test tube.
- 2. Observe the results when the two solutions are combined in the test tube.
- Record your observations for when the two solutions are mixed together and draw and label the contents of the test tube.
- If no precipitate (ppt) is detected, write NR or no reaction.
- 5. Clean your test tubes when finished

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