Name:\_

Date:

# Molecular, Ionic, and Net Ionic Equations

Write down the molecular form of the following equations. Balance and show ionic and net ionic forms.

Molecular equation:

lonic equation:

Net ionic equation:

2) chromium(III) + iron(II) - chromium(III) + iron(II)
nitrate (aq.) + sulfate(aq.) + nitrate(aq.)

Molecular equation:

lonic equation:

Net ionic equation:

3) strontium + potassium -- strontium + potassium bromide(aq.) + sulfate(aq.) -- sulfate(s) + bromide(aq.)

Molecular equation:

lonic equation:

Net ionic equation:

4) copper(II) + potassium copper(II) + potassium carbonate(aq.) + carbonate(s) + sulfate(aq.)

Molecular equation:

Ionic equation:

Net ionic equation:

5) cobalt(III) potassium cobalt potassium bromide(aq.) sulfide(aq.) sulfide(s) bromide(aq.)

Molecular equation:

lonic equation:

Net ionic equation:

# **Practice Net Ionic Equations With Answers**

**Camilla Rothe** 

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