

**Practice Exam AP Chemistry Midterm 1<sup>st</sup> 9 weeks**  
**Section I - Multiple Choice (50%)**

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then blacken the corresponding space on the Scantron. A choice may be once, more than once, or not at all in each set. You must choose 40 out of 50, leave 10 blank. Wrong answers have a 0.25 point penalty, so no guessing!

Questions 1-3

- A. phosphorus
- B. beryllium
- C. sodium
- D. magnesium
- E. sulfur

1. Which of the above is a chalcogen?
2. Which of the above has a nonmetal with an oxidation number of +6?
3. Which of the above forms salts that are all soluble in water?



Questions 4-6

- A.  $\text{CrO}_4^{2-}$
- B.  $\text{SO}_3^{2-}$
- C.  $\text{H}^+$
- D.  $\text{SO}_4^{2-}$
- E.  $\text{Mn}^{2+}$

4. In a reaction between potassium chromate and sodium sulfite in acidified solution, which of the above will serve as the oxidizing agent?
5. Which of the above is added to the reactants side to denote that the reaction is occurring in acidic solution?
6. Forms a yellow color in solution

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**Clifford Lane**



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