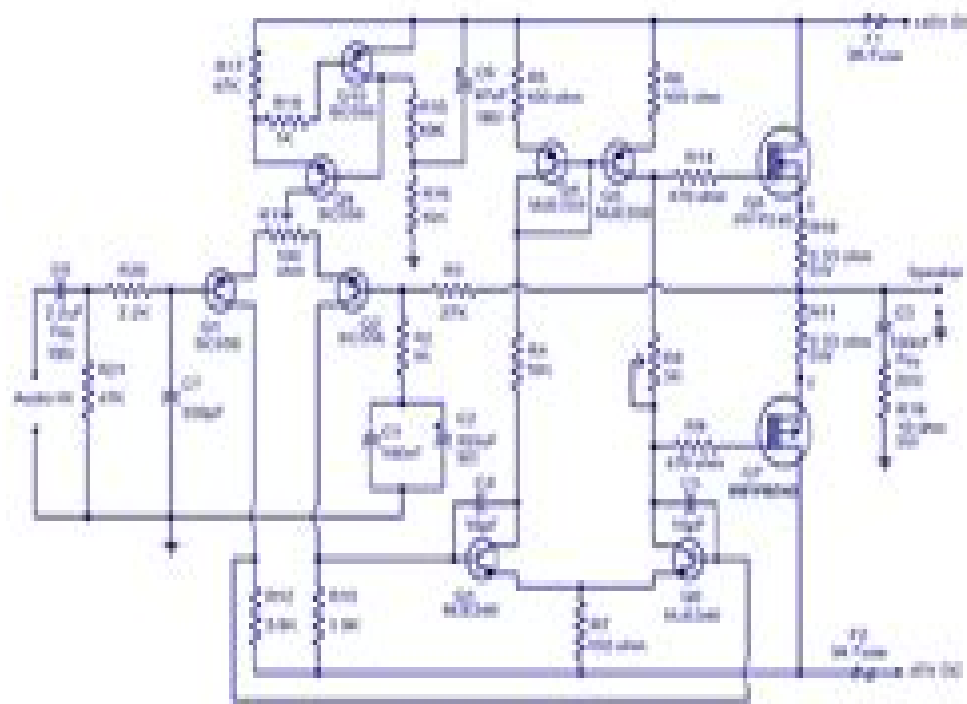


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Siddappa N.Byraredy



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 $\text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$
 $2 \text{NaCl} + \text{F}_2 \rightarrow 2 \text{NaF} + \text{Cl}_2$
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
 $\text{Pb} + \text{OH}^- + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$
 $\text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2\text{SO}_4 + 3 \text{CH}_4$
 $2 \text{O}_2 \rightarrow 1 \text{CO}_2 + 2 \text{H}_2\text{O}$
 $\text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$
 $\text{C}_8\text{H}_{18} + 25 \text{O}_2 \rightarrow 16 \text{CO}_2 + 18 \text{H}_2\text{O}$

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