

More Atomic Structure Practice: Use the periodic table to help you find the answers!

Draw atomic structures and electron dot diagrams for the following atoms: H, He, Li, Be, B, C, O, F, Ne, Na, Al, Si, P, Cl, Ar

Here are the atomic structures: NOTE! Only the electrons in their shells are shown below! The number of protons equals the TOTAL number of electrons. The number of neutrons is found below.

1st Shell	1 H						2 He	
2nd Shell	3 Li	4 Be	5 B	6 C	7 N	8 O	9 F	10 Ne
3rd Shell	11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar

Here are the electron dot diagrams:

I	II			III	IV	V	VI	VII	0
H •									He ••
Li •	• Be •			• B •	• C •	• N •	• O •	• F •	• Ne ••
Na •	• Mg •			• Al •	• Si •	• P •	• S •	• Cl •	• Ar ••
K •	• Ca •			• Ga •	• Ge •	• As •	• Se •	• Br •	• Kr ••
Rb •	• Sr •			• In •	• Sn •	• Sb •	• Te •	• I •	• Xe ••
Cs •	• Ba •			• Tl •	• Pb •	• Bi •	• Po •	• At •	• Rn ••

Metal
 Metalloid
 Nonmetal

How many neutrons are in the following atoms (show your work!): H, He, Li, Be, B, C, O, F, Ne, Na, Al, Si, P, Cl, Ar

Atomic mass (rounded) — Atomic Number = # neutrons

Post Test Atomic Structure And The Periodic Table Key

Padhraic Smyth



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