

Chem 101 Lecture note 2 Possible Multiple Choice Questions and answers

1. Which of the following is NOT a valid postulate of Dalton's Atomic Theory?

- A) Atoms are indestructible.
- B) Atoms of a given element are identical.
- C) Atoms can be converted into other elements.
- D) Atoms combine in simple whole-number ratios to form compounds.

Answer: C

2. J. J. Thomson's model of the atom is often referred to as the:

- A) Solar System Model
- B) Plum Pudding Model
- C) Quantum Model
- D) Nuclear Model

Answer: B

3. What significant discovery is attributed to J. J. Thomson?

- A) Proton
- B) Neutron
- C) Electron
- D) Photon

Answer: C

4. What did Planck propose about energy?

- A) Energy is continuous.
- B) Energy is quantized.
- C) Energy can only exist as waves.
- D) Energy can be created or destroyed.

Answer: B

5. The value of Planck's constant h is:

- A) $6.626 \times 10^{-27} \text{ J.s}$
- B) $6.626 \times 10^{-34} \text{ J.s}$
- C) $1.602 \times 10^{-19} \text{ J.s}$
- D) $9.109 \times 10^{-31} \text{ J.s}$

Answer: B

6. Millikan's Oil Drop Experiment determined the:

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Rachel Sandford



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