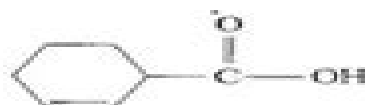


Section A [15 marks]

Answer all questions in this section.

- 1 The structural formula of 2-cyclohexenecarboxylic acid is shown below.



Which is correct about the number of sp^2 and sp^3 hybridised atoms, and σ -bonds in the molecule?

	sp^2	sp^3	σ -bonds
A	3	5	19
B	3	5	8
C	2	4	19
D	2	4	7

- 2 Which compound does **not** correspond correctly to its functional group?

Compound	Functional group
A $C_6H_5CH_2OH$	—OH
B CH_3COOH	$\begin{array}{c} \text{O} \\ \parallel \\ \text{—C—O—} \end{array}$
C CH_3CONH_2	$\begin{array}{c} \text{O} \quad \text{H} \\ \parallel \quad \\ \text{—C—N—H} \end{array}$
D CH_3CHO	$\begin{array}{c} \text{O} \\ \parallel \\ \text{—C—H} \end{array}$

- 3 Alkenes undergo electrophilic addition reaction. The following alkenes react with bromine in CCl_4 to form optically active compounds **except**

A $CH_3CH=CClCH_3$	B $(CH_3)_2C=CHCl$
C $CH_3CH=CHCH_2CH_3$	D $Cl_2C=CHCH_2CH_3$

- 4 Which species can act as an electrophile?

A SO_3	B NH_3	C OH^-	D CN^-
----------	----------	----------	----------

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