## Aqueous Reactions: Neutralization

What chemical species are present in a sample of nitric acid? Make sure to include all atoms, molecules, or ions
present.

Hydrogen ions/Hydronium ions	H*/H3O*
Nitrate ions	NO3
Water molecules	H <sub>2</sub> O

What chemical species are present in a sodium hydroxide solution? Make sure to include all atoms, molecules, or ions present.

Sodium ions	Na*
Hydroxide ions	OH
Water molecules	H <sub>2</sub> O

In the simulation, you mix nitric acid with sodium hydroxide. At the instant that these two solutions are mixed (before
there is time for a reaction to occur), list all species present.

Hydrogen ions/Hydronium ions	H*/H3O*
Nitrate ions	NO <sub>3</sub>
Sodium ions	Na*
Hydroxide ions	OH.
Water molecules	H <sub>2</sub> O

4. Draw a picture to represent the system at the instant that the two solutions are mixed.

# **Neutralization Guided Answers**

**Leonard W. Fine** 

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