



Name _____ Date _____ Block _____

Lab: Redox Half-Reactions

Purpose: To determine the oxidation and reduction half-reactions for simple single-replacement reactions.

Observations

Record your observations below (e.g., gas bubbles or turns black). If nothing occurs, write "No Rxn."

		1	2	3	4	5	6
		ZnSO ₄	H ₂ SO ₄	FeSO ₄	AgNO	CuSO ₄	MgSO ₄
A	Zn	1	2	3	4	5	6
B	Cu	7	8	9	10	11	12
C	Mg	13	14	15	16	17	18
D	Fe	19	20	21	22	23	24

Analysis

For each reaction that took place:

- Write a complete and balanced chemical equation, including phases.
- Assign oxidation numbers for every element
- Write an the oxidation and reduction half-reactions, including phases.

Example	Overall Rxn:	$\text{Ba(s)} + \text{H}_2\text{SO}_4\text{(aq)} \longrightarrow \text{H}_2\text{(g)} + \text{BaSO}_4\text{(aq)}$							
	Oxidation numbers:	0	+1	+6	-2	0	+2	+6	-2
	Oxidation-half Rxn:	$\text{Ba(s)} \longrightarrow \text{Ba}^{2+}\text{(aq)} + 2\text{e}^-$							
	Reduction-half Rxn:	$2\text{H}^+\text{(aq)} + 2\text{e}^- \longrightarrow \text{H}_2\text{(g)}$							

Redox Lab Answer Key

Sally A. Henrie

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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Laboratory Methods in Dynamic Electroanalysis M. Teresa Fernández Abedul, 2019-10-13 *Laboratory Methods in Dynamic Electroanalysis* is a useful guide to introduce analytical chemists and scientists of related disciplines to the world of dynamic electroanalysis using simple and low cost methods. The trend toward decentralization of analysis has made this fascinating field one of the fastest growing branches of analytical chemistry. As electroanalytical devices have moved from conventional electrochemical cells 10-20 mL to current cells e.g. 5-50 mL based on different materials such as paper or polymers that integrate thick or thin film electrodes, interesting strategies have emerged such as the combination of microfluidic cells and biosensing or nanostructuring of electrodes. This book provides detailed easy procedures for dynamic electroanalysis and covers the main trends in electrochemical cells and electrodes including microfluidic electrodes, electrochemical detection in microchip electrophoresis, nanostructuring of electrodes, development of bio-enzymatic, immuno and DNA assays, paper-based electrodes, interdigitated array electrodes, multiplexed analysis and combination with optics. Different strategies and techniques: amperometric, voltammetric and impedimetric are presented in a didactic practice-based way and a bibliography provides readers with additional sources of information. Provides easy-to-implement experiments using low cost simple equipment. Includes laboratory methodologies that utilize both conventional designs and the latest trends in dynamic electroanalysis. Goes beyond the fundamentals covered in other books focusing instead on practical applications of electroanalysis. *Report*, 1972 *Quantitative Analysis Laboratory Manual* Dr. Bupinderpal Singh, Dr. Bikramjit Singh, Mr. Akash Bans, 2025-06-29 *Electroanalytical Techniques in Clinical Chemistry and Laboratory Medicine* Joseph Wang, 1988-10-21 This practical introduction to all the electroanalytical techniques that are used in clinical chemistry and laboratory medicine is the only in-depth treatment of the subject available. The author presents the relevant theory and uses numerous examples to illustrate the scope and possibilities of electroanalysis in the clinical laboratory. The material covered includes the principles and bioanalytical applications of voltammetry and potentiometry, electrochemical biosensors, detectors to flowing streams and in vivo electrochemistry. The book should be useful to those considering the use of electroanalysis in their laboratories and to clinical chemists experienced in electroanalysis. *Working with Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment. The scenarios are drawn from the lives of people who work with chemistry every day, ranging from field ecologists to chemical engineers and include many health professionals as well. **Energy Research Abstracts**, 1993 **Soil Physical Chemistry** Donald L. Sparks, 2018-02-06 *Soil Physical Chemistry* Second Edition takes up where the last edition left off. With comprehensive and contemporary discussions on equilibrium and kinetic aspects of major soil chemical processes and reactions, this excellent text/reference presents new chapters on precipitation/dissolution modeling.

of adsorption reactions at the mineral water interface and the chemistry of humic substances An emphasis is placed on understanding soil chemical reactions from a microscopic point of view and rigorous theoretical developments such as the use of modern in situ surface chemical probes such as x ray adsorption fine structure XAFS Fourier transform infrared FTIR spectroscopies and scanning probe microscopies SPM are discussed **40 Inquiry Exercises for the College Biology**

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Green Chemistry Laboratory Manual for General Chemistry Sally A. Henrie, 2015-03-18. Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts. Providing educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual, this lab manual enables students to see how green chemistry principles can be applied to real world issues. Following a consistent format, each lab experiment includes objectives, prelab questions, and detailed step by step procedures for performing the experiments. Additional questions encourage further research about how green chemistry principles compare with traditional more hazardous experimental methods.

Scientific and Technical Aerospace Reports, 1995 **Glass**, 1986 **Issues in Specialized Chemical and Chemistry Topics: 2013 Edition**, 2013-05-01. Issues in Specialized Chemical and Chemistry Topics 2013 Edition is a ScholarlyEditions book that delivers timely, authoritative and comprehensive information about Magnetic Resonance. The editors have built Issues in Specialized Chemical and Chemistry Topics 2013 Edition on the vast information databases of ScholarlyNews. You can expect the information about Magnetic Resonance in this book to be deeper than what you can access anywhere else, as well as consistently reliable, authoritative, informed and relevant. The content of Issues in Specialized Chemical and Chemistry Topics 2013 Edition has been produced by the world's leading scientists, engineers, analysts, research institutions and companies. All of the content is from peer-reviewed sources and all of it is written, assembled and edited by the editors at ScholarlyEditions and available exclusively from us. You now have a source you can cite with authority, confidence and credibility. More information is available at <http://www.ScholarlyEditions.com>.

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