#### Topics:

- 1. General Chemistry
- 2. Molecular Representations
- 3. Nomenclature
- 4. Alkanes/Cycloalkanes
- 5. Stereoisomers
- 6. Chemical reactivity/mechanism
- 7. SN/E Reactions

#### General Chemistry Questions

- 1. Identify all of the intermolecular forces present in each molecule:
  - a. CH<sub>2</sub>OH
  - b. CH<sub>2</sub>COCH<sub>2</sub>CH<sub>3</sub>
  - c. CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>
- 2. Fill out the blanks in the chart (VESPR Theory)

Hybridization	Geometry	# of Bonds	# of Lone Pairs	Bond Angle
	Tetrahedral	The state of the s	0	The second second
Sp <sup>3</sup>		3	1	
	Bent			109.5
Sp²	3,550	3	0	
		2	1	120
			0	180

- 3. Determine if the following molecules have a dipole moment
  - a. CH<sub>2</sub>CH<sub>3</sub>
  - b. CH3CH2CH(NH<sub>2</sub>)CH<sub>3</sub>
  - c. CH<sub>2</sub>Cl<sub>2</sub>
  - d. H<sub>2</sub>O\*
  - e. HOOst

#### Molecular Representations Questions:

1. Write the condensed name for the following structures:



# **Sample Questions On Chemistry 1exam**

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## Sample Questions On Chemistry 1exam Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "Sample Questions On Chemistry 1exam," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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