

## **Sample character sketch - my friend Liz.**

My friend Liz is the most amazing friend anybody could ask for. We've been through so much together, we're basically like sisters. We met on the first day of school in sixth grade, both of us terrified by the massive size of the middle school. She had the locker right above mine. I told her I didn't know anybody in our class and she said "You do now." We've been friends ever since.

Most boys think Liz is cute. She has long red hair, cascading over her shoulders. She laughs about everything and when she does, you see about a hundred white teeth - so bright, you almost need sunglasses. When she laughs, her eyes grow wide, glowing emerald green. Liz likes to dress kind of skater-ish, in camouflage pants, sweatshirts, and wristbands. But, she's unpredictable, too. Sometimes she'll wear overalls or a fancy dress. She must have three closets full of clothes, because she barely ever wears the same outfit twice.

Liz is the most lively, animated character I've ever known. She's always rushing around, trying to get the latest scoop on everybody. It's like she's in the FBI. Right before she shares important news, Liz tosses back her hair, takes a deep breath, and quickly looks side to side, to be sure the coast is clear. She never says anything mean about people, she just wants to know what's going on.

She always supports me in everything I want to do. Not many girls in our group of friends play sports, but when I told Liz I wanted to go out for basketball, she said "Go for it." Now, she comes to see almost every game I play and cheer me on.

Not only is Liz a tremendous supporter, she also trusts me to give her my honest opinion and to say what I feel. Last year, she thought Mrs. Jones gave her a lower math grade than she deserved. I told her the truth - that Liz handed everything in late and what did she expect? Next marking period, Liz got her work in on time, and pulled off an A-. Thanks to me, she said. Liz is a wonderful listener. She lets me tell her all my problems and she never diminishes the importance of my worries. I can tell she's really listening, too, because she looks directly in my eyes the whole time, like she's trying to see inside my head and figure me out.

We don't always agree on everything, and sometimes we even fight like sisters. But, in the end, we always stick together.

# Sample Of A Character Sketch

**Julia Schneider**



## **Sample Of A Character Sketch:**

## **Sample Of A Character Sketch** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Sample Of A Character Sketch**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall affect readers.

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web solution total mass of the answer 100 g mass of benzene 30 g therefore mass of carbon tet 100 30 g benzene s molar mass  $C_6H_6$  6 12 6 1 g mol 1 78 g mol 1 hence  $C_6H_6$  s number of moles  $\frac{30}{78}$  mol 0 3846 mol molar mass of carbon tet  $CCl_4$  1 12 4 355 154 g mol 1

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web jan 30 2020 ans the molality of solution is 1 mol kg 1 or 1 m example 02 11 11 g of urea  $NH_2CONH_2$  was dissolved in 100 g of water calculate the molarity and molality of the solution given n 14 h 1 c 12 o 16 given mass of solute urea 11 11 g mass of solvent water 100 g 0 1 kg to find molarity of solution solution

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web solution moles of  $CH_3OH$  5 5 32 0 17 mole moles of  $H_2O$  40 18 2 2 moles therefore according to the equation mole fraction of  $CH_3OH$  0 17 2 2 0 17 mole fraction of  $CH_3OH$  0 073 to solve more examples on mole fraction formula and practice more question please visit byju s com

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web you may work out the mole fractions on your own problem 5 what is the mass of a sample of a 0 449 molal kbr that contains 2 92 kg of water 1 molality moles solute divided by kilograms solute 0 449 mol kg x 2 92 kg x 1 31108 mol of kbr 2 moles times molar mass equals grams 1 31108 mol 119 0023 g mol 156 g kbr

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web calculate the mass percent mole fraction molality and molarity of hexane  $C_6H_{14}$  d 0 660 g  $cm^3$  when it is mixed with heptane  $C_7H_{16}$  d 0 684 g  $cm^3$  in a 35 0 55 0 ml ratio assume that the volumes are additive

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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web jan 8 2022 mol % x 100 the sum of all of the mole percents of a mixture equals 100 mole fraction properties and advantages the mole fraction offers advantages over some of the other units of concentration unlike molarity mole fraction is *worksheet mole fraction ms fleming* - Jun 11 2023

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web a derivatization that achieves this goes as molarity moles of solute volume of solution in liter moles molarity x volume in liter mass moles x molar mass numerical answers 1

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web solution a  $3.8 \times 10^{-51}$  grams b  $4.11 \times 10^{-45}$  grams c  $7.31 \times 10^{-25}$  grams 4 7 1 mole practice with guidance exercises is shared under a not declared license and was authored remixed and or curated by libretexts these are homework exercises to accompany the textmap created for chemistry a molecular approach by nivaldo tro

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web what is the mole fraction of licl in the solution 11 8 practice problem acetone c 3 h 6 o has a density of 0 791 g ml and xylene c 8 h 10 has a density of 0 864 g ml a solution was prepared by mixing acetone and xylene calculate the mole fraction of acetone in a solution if 19 0 ml of acetone was dissolved in 98 4 ml of xylene 7 1

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web since the mole fraction of the o 2 is 0 050 it contributes 0 050 or if you will 5 0 of the total pressure thus total pressure is 0 21 0 050 4 2 atm problem 9 a sample of 1 43 g of helium and an unweighed quantity of o 2 are mixed in a flask at room temperature