

SAMPLE PAPER (2023 -24)

CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
 - (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
 - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
 - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
 - (e) SECTION D consists of 2 case - based questions carrying 4 marks each.
 - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
 - (g) All questions are compulsory.
 - (h) Use of log tables and calculators is not allowed.
-

SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

1. Which of the following solutions will have the highest conductivity at 298 K?
- (a) 0.01 M HCl solution
 - (b) 0.1 M HCl solution
 - (c) 0.01 M CH_3COOH solution
 - (d) 0.1 M CH_3COOH solution



Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
 - (b) A = Benzophenone, B = formaldehyde
 - (c) A = Benzaldehyde, B = Acetophenone
 - (d) A = Benzophenone, B = Acetophenone
3. The vitamins which can be stored in our body are:
- (a) Vitamin A, B, D and E
 - (b) Vitamin A, C, D and K
 - (c) Vitamin A, B, C and D
 - (d) Vitamin A, D, E and K
4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{COONa}$ and yellow precipitate of CHI_3 ?
- (a) 3-Methylpent-3-en-2-one
 - (b) 3-Methylbut-2-en- one
 - (c) 2, 3-Dimethylethanone
 - (d) 3-Methylpent-4-one
5. Which of the following is not correct?
- (a) In haloarenes, the electron pairs on halogen atom are in conjugation with π -electrons of the ring.
 - (b) The carbon-magnesium bond is covalent and non-polar in nature .
 - (c) During $\text{S}_{\text{N}}1$ reaction, the carbocation formed in the slow step being sp^2 hybridised is planar.
 - (d) Out of $\text{CH}_2=\text{CH}-\text{Cl}$ and $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$, $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$ is more reactive towards $\text{S}_{\text{N}}1$ reaction

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Table of Contents Sample Question Paper Applied Chemistry

1. Understanding the eBook Sample Question Paper Applied Chemistry
 - The Rise of Digital Reading Sample Question Paper Applied Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Question Paper Applied Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Question Paper Applied Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Question Paper Applied Chemistry

- Personalized Recommendations
 - Sample Question Paper Applied Chemistry User Reviews and Ratings
 - Sample Question Paper Applied Chemistry and Bestseller Lists
5. Accessing Sample Question Paper Applied Chemistry Free and Paid eBooks
 - Sample Question Paper Applied Chemistry Public Domain eBooks
 - Sample Question Paper Applied Chemistry eBook Subscription Services
 - Sample Question Paper Applied Chemistry Budget-Friendly Options
 6. Navigating Sample Question Paper Applied Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Question Paper Applied Chemistry Compatibility with Devices
 - Sample Question Paper Applied Chemistry Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Question Paper Applied Chemistry
 - Highlighting and Note-Taking Sample Question Paper Applied Chemistry
 - Interactive Elements Sample Question Paper Applied Chemistry
 8. Staying Engaged with Sample Question Paper Applied Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Question Paper Applied Chemistry
 9. Balancing eBooks and Physical Books Sample Question Paper Applied Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Question Paper Applied Chemistry
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Sample Question Paper Applied Chemistry
 - Setting Reading Goals Sample Question Paper Applied Chemistry
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Sample Question Paper Applied Chemistry

- Fact-Checking eBook Content of Sample Question Paper Applied Chemistry
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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