

# ICSE Class 10 Chemistry Question Paper Solution 2016

## CHEMISTRY SCIENCE Paper – 2

### Question 1

- (a) Fill in the blanks with the choices given in brackets. [5]
- (i) Metals are good \_\_\_\_\_ (*oxidizing agents / reducing agents*) because they are electron \_\_\_\_\_ (*acceptors/ donors*).
  - (ii) Electrovalent compounds have \_\_\_\_\_ (*high / low*) melting points.
  - (iii) Higher the pH value of a solution, the more \_\_\_\_\_ (*acidic / alkaline*) it is.
  - (iv) \_\_\_\_\_ (*AgCl / PbCl<sub>2</sub>*), a white precipitate is soluble in excess  $\text{NH}_4\text{OH}$ .
  - (v) Conversion of ethene to ethane is an example of \_\_\_\_\_ (*hydration / hydrogenation*).
- (b) Choose the *correct answer* from the options given below: [5]
- (i) An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
    - A. 17
    - B. 11
    - C. 18
    - D. 20
  - (ii) The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
    - A. 1:2
    - B. 1:0.01
    - C. 1:1
    - D. 0.01:1 [Given that H=1, O=16]
  - (iii) The two main metals in *Bronze* are:
    - A. Copper and zinc
    - B. Copper and lead
    - C. Copper and nickel
    - D. Copper and tin
  - (iv) The particles present in *strong electrolytes* are:
    - A. only molecules
    - B. mainly ions

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**Rachel Sandford**



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