

## EASEL ACID AND BASE QUIZ

Multiple Choice: Select the most correct answer. (15 K)

1. Which of the following indicates an acidic solution?

- A) pH = 7.5                      B) pH = 7                      C) pH = 6.5  
D) more than one correct response      E) no correct response

2. Which of the following statements about weak acids is correct?

- A) Weak acids always contain C atoms  
B) A percentage dissociation of 50% is a weak acid  
C) Weak acid molecules readily lose their acidic hydrogen atoms  
D) more than one correct response  
E) none of the above

3. Which of the following equations represents an acid-base neutralization reaction?

- A)  $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$                       B)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$   
C)  $\text{Ba}(\text{OH})_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$       D) more than one correct response  
E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

- A) increases by a factor of 1.5.                      B) increases by a factor of 100.  
C) decreases by a factor of 1.5.                      D) decreases by a factor of 100.

5. The most basic oxides are formed from elements found in the \_\_\_\_\_ region of the periodic table.

- A) right      B) left      C) center      D) top      E) none of the above

6. When  $\text{CaO}(\text{s})$  reacts with water which of the following is true?

- a) The solution contains  $\text{OH}^-_{(\text{aq})}$  and  $\text{Ca}^{2+}_{(\text{aq})}$                       b) The solution contains  $\text{O}^{2-}_{(\text{aq})}$  and  $\text{Ca}^{2+}_{(\text{aq})}$   
c) The solution contains  $\text{O}^{2-}_{(\text{aq})}$ ,  $\text{OH}^-_{(\text{aq})}$ , and  $\text{Ca}^{2+}_{(\text{aq})}$                       d) The solution contains  $\text{CaO}_{(\text{aq})}$

7. A weak base:

- A) partially dissociates in water to give a little  $\text{OH}^-_{(\text{aq})}$   
B) fully reacts with water to form  $\text{OH}^-_{(\text{aq})}$   
C) completely dissociates to form  $\text{OH}^-_{(\text{aq})}$   
D) has a very high pH  
E) contains a group 1 metal and a hydroxide

8. Which of the following is NOT a property of bases?

- A) They react with active metals.                      B) They taste bitter and feel slippery  
C) They have a pH greater than 7.                      D) They turn red litmus blue      E) They neutralize acids.

# Quiz Properties Of Acids And Bases Answers

**William McPherson, William Edwards  
Henderson**



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### **Table of Contents Quiz Properties Of Acids And Bases Answers**

1. Understanding the eBook Quiz Properties Of Acids And Bases Answers
  - The Rise of Digital Reading Quiz Properties Of Acids And Bases Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Quiz Properties Of Acids And Bases Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Quiz Properties Of Acids And Bases Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Quiz Properties Of Acids And Bases Answers
  - Personalized Recommendations
  - Quiz Properties Of Acids And Bases Answers User Reviews and Ratings
  - Quiz Properties Of Acids And Bases Answers and Bestseller Lists

5. Accessing Quiz Properties Of Acids And Bases Answers Free and Paid eBooks
  - Quiz Properties Of Acids And Bases Answers Public Domain eBooks
  - Quiz Properties Of Acids And Bases Answers eBook Subscription Services
  - Quiz Properties Of Acids And Bases Answers Budget-Friendly Options
6. Navigating Quiz Properties Of Acids And Bases Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Quiz Properties Of Acids And Bases Answers Compatibility with Devices
  - Quiz Properties Of Acids And Bases Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Quiz Properties Of Acids And Bases Answers
  - Highlighting and Note-Taking Quiz Properties Of Acids And Bases Answers
  - Interactive Elements Quiz Properties Of Acids And Bases Answers
8. Staying Engaged with Quiz Properties Of Acids And Bases Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Quiz Properties Of Acids And Bases Answers
9. Balancing eBooks and Physical Books Quiz Properties Of Acids And Bases Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Quiz Properties Of Acids And Bases Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Quiz Properties Of Acids And Bases Answers
  - Setting Reading Goals Quiz Properties Of Acids And Bases Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Quiz Properties Of Acids And Bases Answers
  - Fact-Checking eBook Content of Quiz Properties Of Acids And Bases Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning



- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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