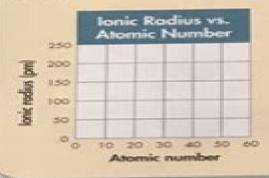
## Quick Lab

Purpose To use a graph to identify period and group trends

#### Materials

- graph paper
- pencil



## Periodic Trends in Ionic Radii

#### Procedure

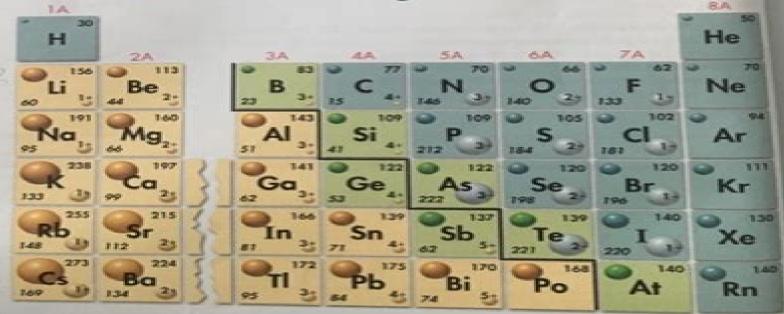
Use the data presented in Figure 6.23 to plot ionic radius versus atomic number.

### Analyze and Conclude

- 1. Compare How does the size change when an atom forms a cation and when an atom forms an anion?
- 2. Describe How do the ionic radii vary within a group of metals? How do they vary within a group of nonmetals?
- 3. Describe What is the shape of a portion of the graph that corresponds to one period?
- 4. Compore and Contrast Is the trend across a period similar or different for Periods 2, 3, 4, and 5?
- 5. Explain Propose explanations for the trends you have described for ionic radii within groups and across periods.

Figure 6.23 Atomic and Ionic Radii Atomic and ionic radii are an indication of the relative size of atoms and ions. The data listed are reported in picometers (pm).





# **Quick Lab Periodic Trends In Ionic Radii Answers**

**LP Steffe** 

Quick Lab Periodic Trends In Ionic Radii Answers:

#### Reviewing Quick Lab Periodic Trends In Ionic Radii Answers: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "Quick Lab Periodic Trends In Ionic Radii Answers," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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