

# Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

12.01

Na

Sodium

24.31

Mg

Magnesium

39.10

K

Potassium

87.62

Rb

Rubidium

132.91

Cs

Cesium

223.02

Fr

Francium

4.00

He

Helium

12.01

B

Boron

12.01

C

Carbon

14.01

N

Nitrogen

16.00

O

Oxygen

18.99

F

Fluorine

20.18

Ne

Neon

26.98

Al

Aluminum

28.09

Si

Silicon

30.97

P

Phosphorus

32.07

S

Sulfur

35.45

Cl

Chlorine

39.96

Ar

Argon

69.72

Ga

Gallium

72.64

Ge

Germanium

74.92

As

Arsenic

78.96

Se

Selenium

81.77

Br

Bromine

83.80

Kr

Krypton

114.82

In

Indium

118.71

Sn

Tin

121.76

Sb

Antimony

127.60

Te

Tellurium

126.91

I

Iodine

131.29

Xe

Xenon

162.57

Tl

Thallium

167.26

Pb

Lead

175.08

Bi

Bismuth

173.04

Po

Polonium

174.93

At

Astatine

175.94

Rn

Radon

14.01

Si

Silicon

14

14.01

28.09

28.09

symbol:

black

solid

blue

liquid

red

gas

white

synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

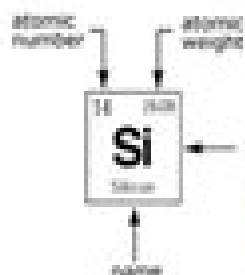
other metals

nonmetals

noble gases

LABS

|   |    |    |    |   |   |   |   |   |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |
|---|----|----|----|---|---|---|---|---|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 1 | 2  | 3  | 4  | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 | 19 | 20 | 21 | 22 | 23 | 24 | 25 | 26 | 27 | 28 | 29 | 30 | 31 | 32 | 33 | 34 | 35 | 36 | 37 | 38 | 39 | 40 | 41 | 42 | 43 | 44 | 45 | 46 | 47 | 48 | 49 | 50 | 51 | 52 | 53 | 54 | 55 | 56 | 57 | 58 | 59 | 60 | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 | 71 | 72 | 73 | 74 | 75 | 76 | 77 | 78 | 79 | 80 | 81 | 82 | 83 | 84 | 85 | 86 | 87 | 88 | 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 | 104 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 | 113 | 114 | 115 | 116 | 117 | 118 |
| H | He | Li | Be | B | C | N | O | F | Ne | Na | Mg | Al | Si | P  | S  | Cl | Ar | K  | Ca | Sc | Ti | V  | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga | Ge | As | Se | Br | Kr | Rb | Sr | Y  | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In | Sn | Sb | Te | I  | Xe | Cs | Ba | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb | Lu | Hf | Ta | W  | Re | Os | Ir | Pt | Au | Hg | Tl | Pb | Bi | Po | At | Rn | Fr | Ra | Ac | Th | Pa | U  | Np | Pu | Am | Cm | Bk | Cf | Es | Fm  | Md  | No  | Lr  |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |



Legend for element states:

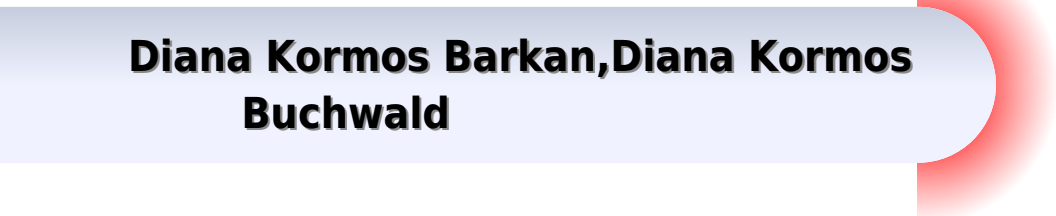
- Black: solid
- Blue: liquid
- Red: gas
- White: synthetically prepared
- Grey: other metals

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

|                     |                          |                       |                        |                       |                       |                        |                       |                         |                         |                      |                          |                       |                         |
|---------------------|--------------------------|-----------------------|------------------------|-----------------------|-----------------------|------------------------|-----------------------|-------------------------|-------------------------|----------------------|--------------------------|-----------------------|-------------------------|
| 58<br>Ce<br>Cerium  | 59<br>Pr<br>Praseodymium | 60<br>Nd<br>Neodymium | 61<br>Pm<br>Promethium | 62<br>Sm<br>Samarium  | 63<br>Eu<br>Europium  | 64<br>Gd<br>Gadolinium | 65<br>Tb<br>Terbium   | 66<br>Dy<br>Dysprosium  | 67<br>Ho<br>Holmium     | 68<br>Er<br>Erbium   | 69<br>Tm<br>Thulium      | 70<br>Yb<br>Ytterbium | 71<br>Lu<br>Lutetium    |
| 90<br>Th<br>Thorium | 91<br>Pa<br>Protactinium | 92<br>U<br>Uranium    | 93<br>Np<br>Neptunium  | 94<br>Pu<br>Plutonium | 95<br>Am<br>Americium | 96<br>Cm<br>Curium     | 97<br>Bk<br>Berkelium | 98<br>Cf<br>Californium | 99<br>Es<br>Einsteinium | 100<br>Fm<br>Fermium | 101<br>Md<br>Mendelevium | 102<br>No<br>Nobelium | 103<br>Lr<br>Lawrencium |

# Physical Science For Dummies

**Diana Kormos Barkan, Diana Kormos  
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