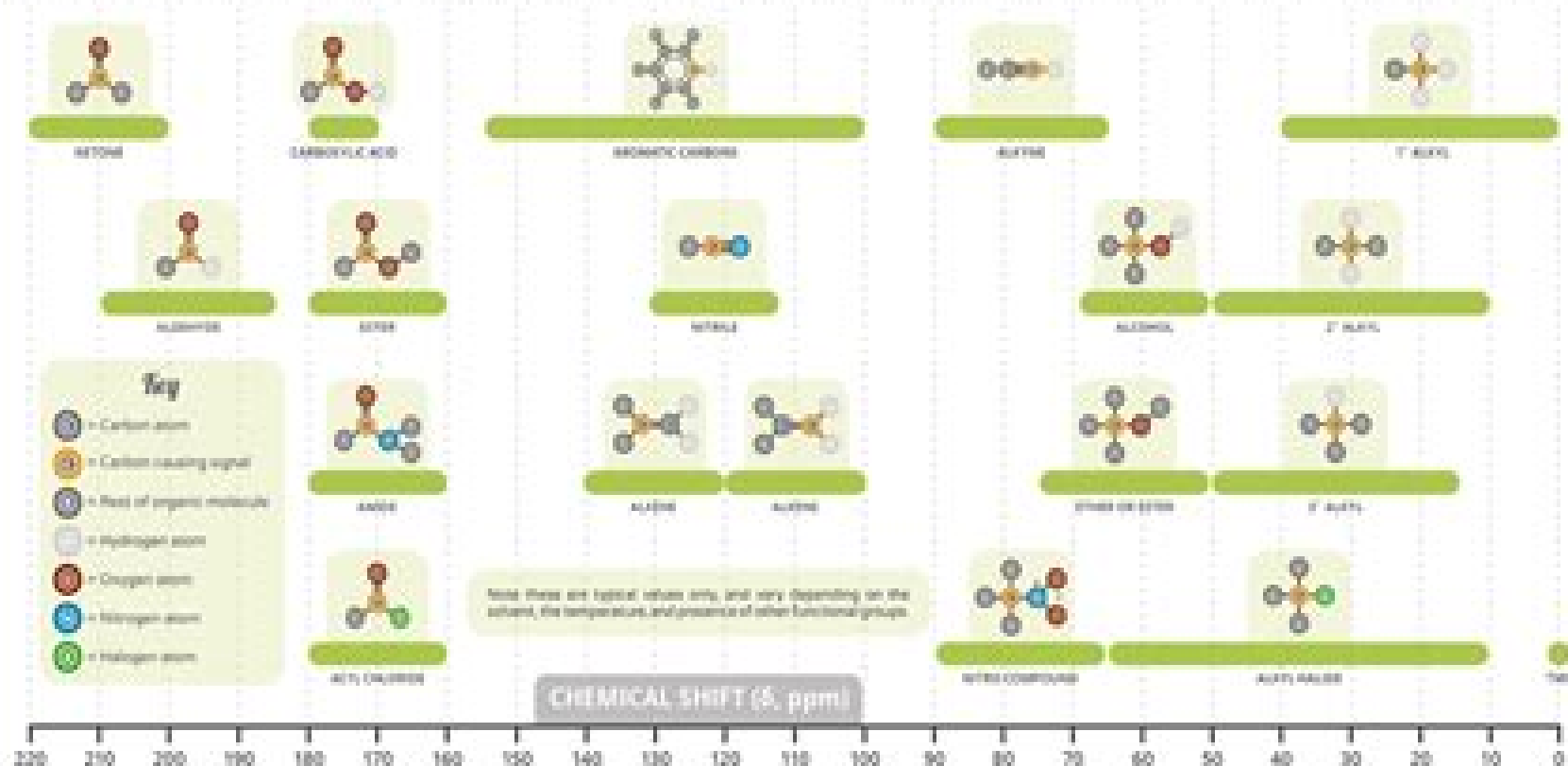


A GUIDE TO ^{13}C NMR CHEMICAL SHIFT VALUES

Nuclear Magnetic Resonance (NMR) is a commonly used technique for organic compound structure determination. In ^{13}C NMR, applying an external magnetic field causes the nuclei spin to flip. The environment of the carbon atom in the molecule affects where the signal is seen on the resultant spectrum.



^{12}C 99% ^{13}C 1%

Only 1% of carbon atoms are carbon-13, atoms which have one more neutron than carbon-12. NMR doesn't work for carbon-12, as its nucleus doesn't have a 'spin'. The frequency required to 'flip' a carbon-13 nucleus is around a quarter of that required to flip an hydrogen nucleus in ^1H -NMR. As the probability of two adjacent carbons in a single molecule being carbon-13 atoms is very low, no splitting of peaks is seen, unlike in ^1H -NMR.



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David Kirk



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Table of Contents Nmr Spectroscopy Guide

1. Understanding the eBook Nmr Spectroscopy Guide
 - The Rise of Digital Reading Nmr Spectroscopy Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Nmr Spectroscopy Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nmr Spectroscopy Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nmr Spectroscopy Guide
 - Personalized Recommendations

- Nmr Spectroscopy Guide User Reviews and Ratings
- Nmr Spectroscopy Guide and Bestseller Lists
- 5. Accessing Nmr Spectroscopy Guide Free and Paid eBooks
 - Nmr Spectroscopy Guide Public Domain eBooks
 - Nmr Spectroscopy Guide eBook Subscription Services
 - Nmr Spectroscopy Guide Budget-Friendly Options
- 6. Navigating Nmr Spectroscopy Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Nmr Spectroscopy Guide Compatibility with Devices
 - Nmr Spectroscopy Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nmr Spectroscopy Guide
 - Highlighting and Note-Taking Nmr Spectroscopy Guide
 - Interactive Elements Nmr Spectroscopy Guide
- 8. Staying Engaged with Nmr Spectroscopy Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nmr Spectroscopy Guide
- 9. Balancing eBooks and Physical Books Nmr Spectroscopy Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nmr Spectroscopy Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nmr Spectroscopy Guide
 - Setting Reading Goals Nmr Spectroscopy Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nmr Spectroscopy Guide
 - Fact-Checking eBook Content of Nmr Spectroscopy Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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