

**SAMSUNG**

# 5.1CH Blu-ray Home Theater System

Model Name : HT-BD1250

Model Code : HT-BD1250T/XAC

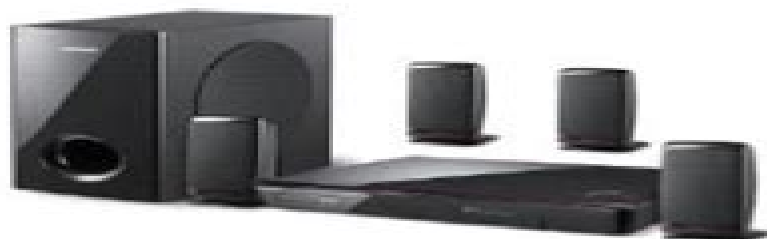
| Speaker   | PS-BD1250  |
|-----------|------------|
| Front     | PS-FBD1250 |
| Center    | PS-CBD1250 |
| Rear      | PS-RBD1250 |
| Subwoofer | PS-WBD1250 |

## ***SERVICE*** Manual

### 5.1CH Blu-ray Home Theater System

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HT-BD1250

# Samsung Ht Bd1250 Bd1250t Service Manual Repair Guide

**Ian Pickup**



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web zero degrees celsius or 273 k is defined as standard temperature and one atmosphere or 760 mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is  $22.4 \text{ l}$

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web at standard temperature and pressure stp the molar volume  $V_m$  is the volume occupied by one mole of a chemical element or a chemical compound it can be calculated by dividing the molar mass  $M$  by mass density  $\rho$  molar gas volume is one mole of any gas at a specific temperature and pressure has a fixed volume molar volume formula

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from gas volume to moles the equality of 1 mol 22.4 l is the basis for the conversion factor example 10.7 l

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web jun 8 2023  $V = 1.00 \text{ mol} \times 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1} \times 273 \text{ K} = 22.4 \text{ L}$  in other words 1 mole of a gas will occupy 22.4 l at stp assuming ideal gas behavior at stp the volume of a gas is only dependent on number of moles of

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