

Practice Paper (Session: 2023-24)

Class: IX

Subject: N. Science(086)

Duration: 3 hours

Maximum Marks: 80

**General Instructions :**

*Read the following instructions very carefully and strictly follow them :*

- (i) This question paper comprises **39** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **five** sections – **A, B, C, D** and **E**.
- (iii) **Section A** – Questions No. **1** to **20** are multiple choice questions. Each question carries **1** mark.
- (iv) **Section B** – Questions No. **21** to **26** are very short answer type questions. Each question carries **2** marks. Answer to these questions should be in the range of **30** to **60** words.
- (v) **Section C** – Questions No. **27** to **33** are short answer type questions. Each question carries **3** marks. Answer to these questions should be in the range of **50** to **80** words.
- (vi) **Section D** – Questions No. **34** to **36** are long answer type questions. Each question carries **5** marks. Answer to these questions should be in the range of **80** to **120** words.
- (vii) **Section E** – Questions No. **37** to **39** are of 3 source-based / case-based units of assessment carrying **4** marks each with sub-parts.
- (viii) There is no overall choice. However, an internal choice has been provided in some sections. Only one of the alternatives has to be attempted in such questions.

**SECTION-A**

Q. No.	Questions	Marks
1	The similarity between solid, liquid and gas matter is that, they: a) occupy space                      b) have mass                      c) both a and b                      d) have fixed shape	1
2	When 20 g of potassium permanganate is dissolved in 1000ml of water, there is no increase in volume. This observation indicates that: a) particles of water have spaces between them into which potassium permanganate particles fit. b) particles of water attract each other. c) particle of potassium permanganate attracts each other. d) particles are continuously moving.	1
3	Which of the following is not a Liquid-gas Aerosol: a) mist                      b) clouds                      c) smoke                      d) fog	1
4	The ions present in $\text{Na}_2\text{CO}_3$ are:	1

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