

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte? a) KCl b) CuCl_2 c) AlCl_3 d) MgCl_2
2	The ionic strength of a solution of molality m_i and valency of the ion Z_i is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$
3	Which of the following is an example of electrolyte concentration cell with transference? a) $\text{Cu} \text{CuSO}_4 \text{AgNO}_3 \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)} \text{AgCl}_{(s)} \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)}(a_1) \text{AgNO}_3(a_2) \text{Ag}$ d) $\text{Ag} - \text{AgCl} \text{NaCl}(a_1) \text{NaCl}(a_2) \text{AgCl} \cdot \text{Ag}$
4	A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$
6	The activity 'a' is related to the activity coefficient ' γ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$

Physical Chemistry Multiple Choice Questions

RJ Shavelson



Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25

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The Pearson Guide to Physical Chemistry for the IIT JEE Singhal, **A Textbook of Physical Chemistry, 6th Edition** Sharma K.K. & Sharma L.K., 2016 *A Textbook of Physical Chemistry* *Atkins' Physical Chemistry* Peter William Atkins, Julio De Paula, James Keeler, 2023 Revised edition of Atkins Physical chemistry Peter Atkins Julio de Paula James Keeler Eleventh edition 2018 **Elements of Physical Chemistry** Peter Atkins, Julio de Paula, 2013 *Elements of Physical Chemistry* has been carefully crafted to help students increase their confidence when using physics and mathematics to answer fundamental questions about the structure of molecules how chemical reactions take place and why materials behave the way they do

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[AQA A-level Year 2 Chemistry Student Guide: Physical chemistry 2](#) Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills

Provides the content for generating individual revision notes **Pharmaceutical Physical Chemistry: Theory and Practices** S. K. Bhasin, Physical chemistry is a compulsory paper offered to all the students of pharmacy There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses Pharmaceutical Physical Chemistry Theory and Practices has been designed considering their requirements laid down by AICTE and other premier institutes universities Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy AQA AS/A Level Year 1 Chemistry Student Guide: Physical chemistry 1 Alyn G. McFarland, Nora Henry, 2015-10-09 Exam Board AQA Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps you identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes Multiple Choice Questions in A-level Chemistry B. P. Bourne, 1969 *Essentials of Physical Chemistry 28th Edition* Bahl Arun/ Bahl B.S. & Tuli G.D., 2022-03 *Essentials of Physical Chemistry* is a classic textbook on the subject explaining fundamentals concepts with discussions illustrations and exercises With clear explanation systematic presentation and scientific accuracy the book not only helps the students clear misconceptions about the basic concepts but also enhances students ability to analyse and systematically solve problems This bestseller is primarily designed for B Sc students and would equally be useful for the aspirants of medical and engineering entrance examinations

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