

|       |       |                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          |     |
|-------|-------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----|
|       |       | <p>gerangskik <input checked="" type="checkbox"/> is om die hoeke tussen hulle te maksimeer. <input checked="" type="checkbox"/> Strukturele elektronpare is bindingspare plus alleenpare.</p> <p>OF</p> <p>Valensdop-elektronpaarafstotingsmodel is 'n model om die vorms van molekules te voorspel <input checked="" type="checkbox"/> waarin die elektronpaar vanaf die buitenste dop <input checked="" type="checkbox"/> van 'n verwysingsatoom rondom hierdie atoom gerangskik is om die afstoting tussen hulle te minimaliseer. <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                       |     |
| 1.2.4 |       | <p>The molecule is a tetrahedron <input checked="" type="checkbox"/></p> <p>Die molekule is 'n tetraëder <input checked="" type="checkbox"/></p> <p>1 mark/punt</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | LO2 |
| 1.2.5 |       |  <p>Name trigonal pyramid <input checked="" type="checkbox"/></p> <p>Noem trigonale piramidele <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                            | LO4 |
| 1.3   | 1.3.1 | <p>The electron pair is equally shared between the two Cl atoms <input checked="" type="checkbox"/> and the resulting Cl<sub>2</sub> molecule has no degree of polarity. <input checked="" type="checkbox"/></p> <p>Die elektronpaar word ewe veel gedeel tussen die twee Cl-atome <input checked="" type="checkbox"/> en die resulterende Cl<sub>2</sub>-molekule het geen mate van polariteit nie <input checked="" type="checkbox"/>.</p> <p>2 marks/punte</p>                                                                                                                                                                                                                                                                                                                        | LO2 |
|       | 1.3.2 | <p>The tendency <input checked="" type="checkbox"/> of an atom to pull the bonded pair <input checked="" type="checkbox"/> of electrons towards itself. <input checked="" type="checkbox"/></p> <p>Die neiging <input checked="" type="checkbox"/> van 'n atoom om die gebinde elektronpare <input checked="" type="checkbox"/> na homself te trek <input checked="" type="checkbox"/>.</p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                         | LO2 |
|       | 1.3.3 | <p>Electronegativity of Na = 6    Electronegativity of Cl = 3<br/>Difference is 3 <input checked="" type="checkbox"/></p> <p>With a difference of greater than 2.1 <input checked="" type="checkbox"/> electron transfer will take place and the bond <input checked="" type="checkbox"/> is ionic. <input checked="" type="checkbox"/> (value of 2.1 does not need to be given for this mark)</p> <p>Elektronegatiwiteit van Na = 6    Elektronegatiwiteit van Cl = 3<br/>Verskil is 3 <input checked="" type="checkbox"/></p> <p>Met 'n verskil van meer as 2.1 <input checked="" type="checkbox"/> sal elektroneoordrag plaasvind en die binding is ionies. <input checked="" type="checkbox"/> (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie)</p> <p>4 marks/punte</p> | LO2 |

# Physical Science Control Test February 2014

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