Density Worksheet Physical Science D=m/V

Substance	Density (g/cm²)	Substance	Density (g/cm²)
Osygen	0.00933	Aluminam	9.70
Hydrogen	0.000084	bron	7.87
Ethanol	0.785	Copper	8.96
Benzene	0.880	Silver	30.5
Water	1.000	Lead	31-34
Magnesium	3.74	Mercury	13.6
Salt (sodium chloride)	2.16	Gold	19.32

 The ratio of an object's mass to its ______ is called the devisity of the object.

 A kilogram of lead occupies a much smaller volume than a kilogram of water, because has a much higher density.

 For the masses and volumes indicated, calculate the density in grams per cubic centimeters.

a. mass = 453 g; volume = 225 cm3

b. mass = 5.0 g; volume = 10.0 cm3

c. mass = 26.1 g; volume = 2.0 mL

4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.

A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

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