FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

| - | Homologous Structure of functional | | Example c | of a compound |
|---|------------------------------------|-------------------------------|------------------|--------------------|
| | Homologous series | group | Name | Structural formula |
| | alkanes | Only C-H and C-C single bonds | ethane , | н |
| | alkenes |)c=c(| ethene |)c=< |
| I | alkynes | -c≡c- | ethyne | н-с=с-н |
| | haloalkanes (alkyl halides) | | bromoethane | н—Ё—Ё—вг |
| | alcohols (alkanols) | -¢-о-н | ethanol | H |
| | aldehydes | н_ | ethanal | н—Ё—Н |
| Ī | ketones | _!_%_!_ | proper-2-one | |
| Ī | carboxylic acids | —ё—о-н | ethanoic acid | н-ё-ё-о-н |
| Ī | esters | | methyl ethancate | H |

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon chain.

What and where are the substituents?

parent How many carbons?

Suffix

What is the functional group/homologous series?

di - 2 tri - 3 telra - 4

| Number of carbon atoms | Parent name | Number of carbon atoms | Parent name |
|---------------------------|----------------|---------------------------|----------------|
| 1 | meth | 5 | pent |
| 2 | eth | 6 | hex |
| 3 | prop | 7 | hept |
| 4 | but | 8 | oct |

prefix, suffix and parent!

Page 4 of 41 Tetra – Chloro – melta

Physical Science Organic Compounds Study Guide

Christian Drosten

Physical Science Organic Compounds Study Guide:

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