

**QUESTION 10 (Start on a new page.)**

The balanced chemical equation for the EXTRACTION of gold from its ore is as follows:



- 10.1 State ONE disadvantage of using cyanide ( $\text{CN}^-$ ) in the extraction of gold. (1)
- 10.2 Will the final solution of the extraction process be ACIDIC or BASIC (ALKALINE)? Give a reason for the answer. (2)
- 10.3 Determine the oxidation number of gold in  $\text{NaAu(CN)}_2$ . (1)
- 10.4 Write down the FORMULA of the reducing agent in the reaction above. (1)

Zinc powder is now used to PRECIPITATE the gold.

The balanced equation for the reaction is:



- 10.5 Does zinc undergo OXIDATION or REDUCTION during the precipitation reaction? (1)
- 10.6 Write down a half-reaction to support the answer to QUESTION 10.5. (2)
- 10.7 Calculate the percentage of gold in  $\text{NaAu(CN)}_2$ . (2)
- [10]**

**TOTAL: 150**

# Physical Science P2 Caps Grade1dbe November 2014

**Rachel S Tattersall**



**Physical Science P2 Caps Grade1dbe November 2014:**

The book delves into Physical Science P2 Caps Grade1dbe November 2014. Physical Science P2 Caps Grade1dbe November 2014 is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Science P2 Caps Grade1dbe November 2014, encompassing both the fundamentals and more intricate discussions.

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    - Chapter 3: Physical Science P2 Caps Grade1dbe November 2014 in Everyday Life
    - Chapter 4: Physical Science P2 Caps Grade1dbe November 2014 in Specific Contexts
    - Chapter 5: Conclusion
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