

|       |       |                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          |     |
|-------|-------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----|
|       |       | <p>gerangskik <input checked="" type="checkbox"/> is om die hoeke tussen hulle te maksimeer. <input checked="" type="checkbox"/> Strukturele elektronpare is bindingspare plus alleenpare.</p> <p>OF</p> <p>Valensdop-elektronpaarafstotingsmodel is 'n model om die vorms van molekules te voorspel <input checked="" type="checkbox"/> waarin die elektronpaar vanaf die buitenste dop <input checked="" type="checkbox"/> van 'n verwysingsatoom rondom hierdie atoom gerangskik is om die afstoting tussen hulle te minimaliseer. <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                       |     |
| 1.2.4 |       | <p>The molecule is a tetrahedron <input checked="" type="checkbox"/></p> <p>Die molekule is 'n tetraëder <input checked="" type="checkbox"/></p> <p>1 mark/punt</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | LO2 |
| 1.2.5 |       |  <p>Name trigonal pyramid <input checked="" type="checkbox"/></p> <p>Noem trigonale piramidele <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                            | LO4 |
| 1.3   | 1.3.1 | <p>The electron pair is equally shared between the two Cl atoms <input checked="" type="checkbox"/> and the resulting Cl<sub>2</sub> molecule has no degree of polarity. <input checked="" type="checkbox"/></p> <p>Die elektronpaar word ewe veel gedeel tussen die twee Cl-atome <input checked="" type="checkbox"/> en die resulterende Cl<sub>2</sub>-molekule het geen mate van polariteit nie <input checked="" type="checkbox"/>.</p> <p>2 marks/punte</p>                                                                                                                                                                                                                                                                                                                        | LO2 |
|       | 1.3.2 | <p>The tendency <input checked="" type="checkbox"/> of an atom to pull the bonded pair <input checked="" type="checkbox"/> of electrons towards itself. <input checked="" type="checkbox"/></p> <p>Die neiging <input checked="" type="checkbox"/> van 'n atoom om die gebinde elektronpare <input checked="" type="checkbox"/> na homself te trek <input checked="" type="checkbox"/>.</p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                         | LO2 |
|       | 1.3.3 | <p>Electronegativity of Na = 6    Electronegativity of Cl = 3<br/>Difference is 3 <input checked="" type="checkbox"/></p> <p>With a difference of greater than 2.1 <input checked="" type="checkbox"/> electron transfer will take place and the bond <input checked="" type="checkbox"/> is ionic. <input checked="" type="checkbox"/> (value of 2.1 does not need to be given for this mark)</p> <p>Elektronegatiwiteit van Na = 6    Elektronegatiwiteit van Cl = 3<br/>Verskil is 3 <input checked="" type="checkbox"/></p> <p>Met 'n verskil van meer as 2.1 <input checked="" type="checkbox"/> sal elektroneoordrag plaasvind en die binding is ionies. <input checked="" type="checkbox"/> (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie)</p> <p>4 marks/punte</p> | LO2 |

# Physical Science Memo2013 Control Grade 1september

**Joacim Rocklöv**



**Physical Science Memo2013 Control Grade 1september:**

## Decoding **Physical Science Memo2013 Control Grade 1september**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science Memo2013 Control Grade 1september**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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