Nomenclature Worksheet 2: Simple Binary Ionic Compounds

Please complete the following table:

| Name of Ionic Compound | Formula of Ionic Compound |
|------------------------|--------------------------------|
| Sodium bromide | |
| Calcium chloride | |
| Magnesium sulfide | |
| Aluminum oxide | |
| 5. Lithium phosphide | |
| 6. Cesium nitride | |
| 7. Potassium iodide | |
| Barium fluoride | |
| 9. Rubidium nitride | |
| 10. Barium oxide | |
| 11. | K ₂ O |
| 12. | Mgb |
| 13. | AICl ₃ |
| 14. | CaBr ₂ |
| 15. | Na ₃ N |
| 16. | LiF |
| 17. | Ba ₃ P ₂ |
| 18. | Cs ₂ S |
| 19. | SrF ₂ |
| 20. | NaCl |

Nomenclature Worksheet 2 Answer Key

Judd E. Hollander

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