

The questions in the test bank cover the concepts from the lessons in Chapter 2. The types of questions include multiple choice, true/false, fill-in-the-blank, and short answer.

Chapter II : Atoms, Molecules, and Ions

- Atomic Structure

- The Iron(II) ion has:
 - 28 electrons and a charge of 3+
 - 24 electrons and a charge of 3-
 - 26 electrons and a charge of 2+
 - 24 electrons and a charge of 2+
 - 26 electrons and a charge of 3+
- Determine the number of *electrons* , *protons* , and *neutrons* of $^{56}_{26}\text{Fe}^{3+}$ and $^{204}_{81}\text{Tl}$?
- A neutral atom of the element potassium $^{39}_{19}\text{K}$ has :
 - 19 e, 19 p, 21 n
 - 40 e, 40 p, 19 n
 - 19 e, 21 p, 19 n
 - 40 e, 19 p, 21 n
 - 21 e, 19 p, 19 n
- The number of *neutrons* in an atom of $^{127}_{53}\text{I}$ is :
 - 53
 - 74
 - 127
 - 180
- An atom of the isotope chlorine-39 consists of how many *protons*, *neutrons*, and *electrons*? (p = proton, n= neutron, e = electron)
 - 17 p, 17 n, 22 e
 - 39 p, 38 n, 39 e
 - 22 p, 17 n, 17 e
 - 17 p, 39 n, 17 e
 - 17 p, 22 n, 17 e
- How many **protons** and **electrons** are present in a $^{40}_{20}\text{Ca}^{2+}$ ion?
 - 18 p, 20 e
 - 20 p, 18 e
 - 20 p, 21 e
 - 20 p, 19 e
 - 22 p, 20 e
- The elements in a *row* of the *periodic table* are known as:
 - metalloids
 - noble gases
 - a group
 - nonmetals
 - a period
- True or False :

() The number of electrons in natural atom equal to the number of neutrons

- Isotopes

- Two *isotopes* of an element *differ* in their:
 - atomic mass
 - atomic number
 - number of protons
 - number of electrons
- For the following data about unknown element :

92.21%	(27.977)	4.700%	(28.976)	3.090%	(29.974)
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Identify the element ?

Science Focus Test Chapter Atoms

Jin-Ying Zhang



Science Focus Test Chapter Atoms:

Decoding **Science Focus Test Chapter Atoms**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Science Focus Test Chapter Atoms**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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