

1. Identify the phrases that generally apply to molecular compounds,

- | | |
|---|--|
| ☐ A c. Use covalent bonds | ☐ B a. Are made of non-metals chemically joined together. |
| ☐ C d. Use ionic bonds | ☐ D e. Have high boiling points |
| ☐ E b. Have low melting points | |

2. The attraction of the free-floating valence electrons for the positively charged metal ions.

- | | |
|---|--|
| ☐ A Ionic bond | ☐ B Metallic Bond |
| ☐ C London dispersion forces | ☐ D Covalent Bond |

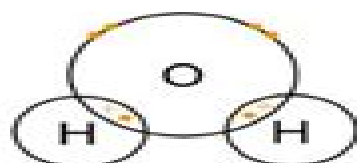
3. Which of the following elements will form a similar ion?

- | | |
|---------------------------------------|--------------------------------------|
| ☐ A Be, Ca, Ba | ☐ B Na, K, Li |
| ☐ C Cl, Ar, Br | ☐ D F, Cl, Br |

4. Which of the following is true about ionic bonding

- | | |
|---|---|
| ☐ A It is weaker than the covalent bond | ☐ B Electrons are donated from one element to another |
| ☐ C Is made by the attraction between a cation and an anion. | ☐ D It is mostly formed between elements from groups IA, IIA, IIIA and a nonmetal. |

5. What type of bond is formed between the hydrogen and oxygen atoms in this image?



- | | |
|--|---|
| ☐ A Metallic bond | ☐ B London Dispersion Forces |
|--|---|

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CL Gary



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