

6.4 Matter Study Guide

Vocabulary:

atom- the basic building block of all matter

matter- anything that has mass and takes up space

period- the horizontal rows on the periodic table

group- the vertical rows on the periodic table

electron- positively charged particles inside the nucleus

electron- particle is the most energetic and moves rapidly around the outside of the nucleus

nucleus- center of an atom; contains protons and neutrons

atomic number- how many protons are in an atom

molecule- atoms that are formed by two nonmetals

element- a substance cannot be broken down chemically into other substances

compound- consists of two or more substances that are all chemically combined

mixture- a substance made of two or more materials that can be separated back into its original materials

solution- a mixture that forms when one substance dissolves another

chemical bond- the force that holds atoms together

chemical change- a change in matter that produces a new substance

physical change- breaking or mixing substances

chemical equation- a shorter, easier way to show chemical reactions, using symbols instead of words



		Periodic Table of the Elements																		
Group	1	2															16	17	18	
Period	1	H																	He	
	2	Li	Be	B	C	N	O	F	Ne									Ar	Kr	
	3	Na	Mg	Al	Si	P	S	Cl	Ar									Xe	Rn	
	4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
	5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
	6	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Rn	
	7	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

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