

For questions 5 to 9, read each question carefully. Decide which of the choices best completes the statement or answers the question. Place your answer in the blank space given.

1

1

5. Iron oxide ($\text{FeO}_{(s)}$) separated into solid iron ($\text{Fe}_{(s)}$) and oxygen ($\text{O}_{2(g)}$) is an example of a

- A. simple composition reaction
- B. simple decomposition reaction
- C. combustion reaction
- D. neutralization reaction

1

1

6. In a simple composition reaction,

- A. one substance combines with another to form a salt and water
- B. two or more elements combine to form a compound
- C. a compound is broken down into two or more elements
- D. water is separated into hydrogen gas and oxygen gas

1

1

7. $\text{fuel} + \text{oxygen} \rightarrow \text{carbon dioxide} + \text{water vapour} + \text{energy}$

This word equation is an example of a

- A. simple composition reaction
- B. simple decomposition reaction
- C. combustion reaction
- D. neutralization reaction

1

1

8. $\text{hydrochloric acid} + \text{magnesium hydroxide} \rightarrow \text{magnesium chloride} + \text{water}$

This word equation is an example of a

- A. simple composition reaction
- B. simple decomposition reaction
- C. combustion reaction
- D. neutralization reaction

1

1

9. Which type of reaction would most likely be endothermic?

- A. a simple composition reaction
- B. a simple decomposition reaction
- C. a combustion reaction
- D. a neutralization reaction

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IM Harris



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