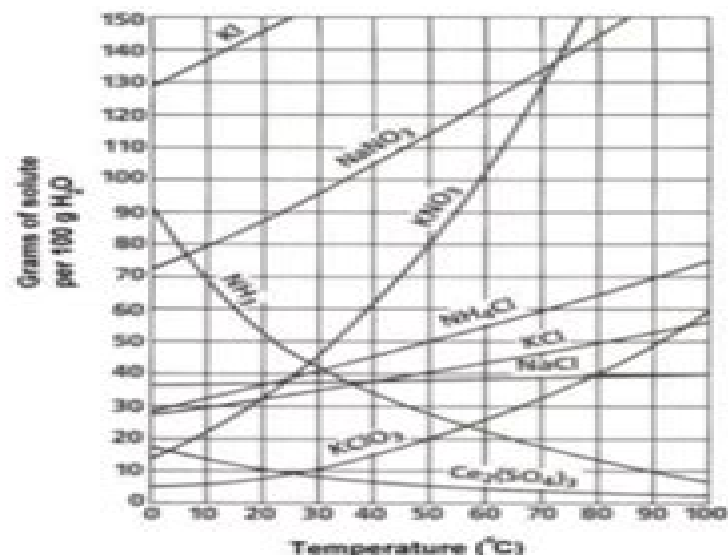


Chemistry I-H
Solubility Curves and Solutions Review Sheet
Solution Set

1. Define solubility: **The maximum amount of solute that dissolves in a fixed quantity of a particular solvent at a specified temperature.**

Use the solubility curve below to answer questions #2 - #15. Be sure to note the units on the axes of the graph.



2. In general, how does temperature affect solubility?
The higher the temp. the more substance dissolves
3. **KClO₃** Which compound is least soluble at 10°C?
4. **50 g** How many grams of KCl can be dissolved in 100g of water at 80°C?
5. **40 g** How many grams of NaCl can be dissolved in 100g of water at 90°C?
6. **about 60 g** At 40°C, how much KNO₃ can be dissolved in 100g of water?
7. **NaCl** Which compound shows the least amount of change in solubility from 0°C-100°C?
8. **unsaturated** At 30°C, 90g of NaNO₃ is dissolved in 100g of water. Is this solution saturated or unsaturated?
9. **D** At 60 C, 72 g of NH₄Cl are dissolved in 100g of water. This solution is considered to be:
A) saturated; B) unsaturated; C) supersaturated; D) saturated with some left undissolved
10. **30 g** A saturated solution of KClO₃ is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?
11. **20 g** A saturated solution of NH₄Cl is formed from one hundred grams of water. If the saturated solution is cooled from 80°C to 40°C, how many grams of precipitate are formed?
12. **E** Which compounds show a *decrease* in solubility from 0°C-100°C?
A) potassium chlorate B) cerium sulfate C) sodium nitrate D) ammonia E) Both B and D
13. **C** Which compound is the most soluble at 10°C?
A) Ce₂(SO₄)₃ B) NaCl C) KCl D) KClO₃ E) NH₃
14. **C** A chemist adds 120 grams of KCl into a test tube which contains 100 grams of water. The temperature of the water is 80°C. How much of the solute will be left undissolved at the bottom of the test tube?
A) 50 grams B) 60 grams C) 70 grams D) 80 grams E) 56 grams

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Understanding Chemistry Herman J. Bruckman, Allan Cruickshanks, 1988 *Catalogue American International College*, 1912 Nuclear Science Abstracts, 1974 Resources in Education, 1984 *Breaking Boundaries: Pioneering Sustainable Solutions Through Materials and Technology* Mir Waqas Alam, 2025-01-24 This book explores sustainable innovation by delving into advanced materials science and technology Each chapter reveals the transformative potential of sustainable solutions from groundbreaking advancements in nanomaterials to eco friendly manufacturing practices This book offers a captivating glimpse into the potential future of sustainability appealing to experienced researchers budding innovators and those with a general interest in the topic Also this book provides valuable insights into recent developments in materials science and technology catering to academics engineers and policymakers It aims to promote collaboration across many disciplines and encourage innovation to speed up the development of sustainable solutions that will have a long lasting positive effect on future generations *Chemical News and Journal of Physical Science*, 1909 *The Chemical News and Journal of Industrial Science*, 1913 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1976 Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text Wastes: Solutions, Treatments and Opportunities III Candida Vilarinho, Fernando Castro, Margarida Gonçalves, Ana Luísa Fernando, 2019-08-08 Wastes Solutions Treatments and Opportunities III contains selected papers presented at the 5th edition of the International Conference Wastes Solutions Treatments and Opportunities that took place on 3 6 September 2019 in Costa da Caparica Portugal The Wastes conference which takes place biennially is a prime forum for sharing innovation technological development and sustainable solutions for the waste management and recycling sectors around the world counting with the participation of experts from academia and industry The papers included in this book cover a wide range of topics including Wastes as construction materials Wastes as fuels Waste treatment technologies MSW management Recycling of wastes and materials recovery Environmental economic and social aspects in waste management Life cycle assessment Circular economy and wastes refineries Logistics policies regulatory constraints and markets in waste management *Chemical News and Journal of Industrial Science*, 1903 **The Chemical News and Journal of Physical Science**, 1903 *Journal of Pharmaceutical Sciences*, 1922 Vols for 1912 45 include proceedings of the association s annual meeting **Engineering Separations Unit Operations for Nuclear Processing** Reid Peterson, 2019-11-15 Engineering Separations Unit Operations for Nuclear Processing provides insight into the fundamentals of separations in nuclear materials processing not covered in typical texts This book integrates fuel cycle and waste processing into a single coherent approach demonstrating that the principles from one field can and should be

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1. Understanding the eBook Science Chemistry Unit Solutions Review
 - The Rise of Digital Reading Science Chemistry Unit Solutions Review
 - Advantages of eBooks Over Traditional Books
2. Identifying Science Chemistry Unit Solutions Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Science Chemistry Unit Solutions Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from Science Chemistry Unit Solutions Review
 - Personalized Recommendations
 - Science Chemistry Unit Solutions Review User Reviews and Ratings
 - Science Chemistry Unit Solutions Review and Bestseller Lists
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 - Highlighting and Note-Taking Science Chemistry Unit Solutions Review
 - Interactive Elements Science Chemistry Unit Solutions Review
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Science Chemistry Unit Solutions Review
 9. Balancing eBooks and Physical Books Science Chemistry Unit Solutions Review
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Science Chemistry Unit Solutions Review
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Science Chemistry Unit Solutions Review
 - Setting Reading Goals Science Chemistry Unit Solutions Review
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Science Chemistry Unit Solutions Review
 - Fact-Checking eBook Content of Science Chemistry Unit Solutions Review
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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