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CHEMISTRY COACHING CIRCLE

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INORGANIC CHEMISTRY TEST - 1

READ INSTRUCTIONS CAREFULLY

1. The test is of 1 hour duration.
2. The maximum marks are 150.
3. This test consists of 50 questions. (Negative Marking)
4. For each question you will be awarded 3 marks if you have darkened only the bubble corresponding to the correct answer and zero mark if no bubbles are darkened. Minus one (-1) mark will be awarded for wrong answer.

1. Among the following, the species that is both paramagnetic and coloured is:

- a. $[\text{MnO}_4]^{2-}$ b. $[\text{TiCl}_6]^{2-}$ c. $[\text{VO}_4]^{3-}$ d. CrO_2Cl_2

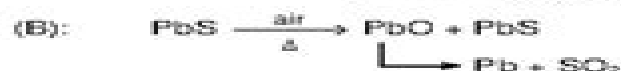
2. Which of the following increasing order of oxidizing power is correct for the following species?
 VO_2^+ , MnO_4^- , $\text{Cr}_2\text{O}_7^{2-}$

- a. $\text{VO}_2^+ < \text{Cr}_2\text{O}_7^{2-} < \text{MnO}_4^-$ b. $\text{VO}_2^+ < \text{MnO}_4^- < \text{Cr}_2\text{O}_7^{2-}$
c. $\text{Cr}_2\text{O}_7^{2-} < \text{VO}_2^+ < \text{MnO}_4^-$ d. $\text{Cr}_2\text{O}_7^{2-} < \text{MnO}_4^- < \text{VO}_2^+$

3. Select the correct statement.

- a. In the decomposition of an oxide into oxygen and gaseous metal, entropy increases.
b. Decomposition of an oxide is an endothermic change.
c. To make ΔG° negative, temperature should be high enough so that $T\Delta S^\circ > \Delta H^\circ$.
d. All are correct statements.

4. Main source of lead is galena (PbS). It is converted to Pb by:



Self – reduction process is:

- a. A b. B c. Both d. none

5. Which of the following statements is incorrect?

- a. Cassiterite ore of tin contains the impurities of Wol-framite which are separated by electromagnetic separator.
b. Tin metal is obtained by the carbon reduction of black indioxide
c. In the extraction of lead from galena, the roasting and self-reduction are carried out in the same furnace at different temperature.
d. Reduction agent of haematite in blast-furnace is coke in upper part and CO in lower part of furnace.

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