

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_6H_{12}O_6$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

ionic $Ca^{2+}Cl_2^{-}$ (ionic)

Formula represents the proportions of _____ of
positive and negative ions in one formula
unit.

Atomic Mass

The proportions that atoms combine through
ions together.

Mass

$100g$ mass electrons

mass proportions charged
ions

Subscript

mass electrons

mass proportions charged

mass

Note Taking Guide Episode 601 Chemical Formula Key

Michael Brown



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