

### Quantum Theory

- a. Energy is quantized in  $\frac{h\nu}{2\pi}$  and  $\frac{h\nu}{2\pi}$  is the minimum energy
- b. Small packets of energy is called as photons
- c. In quantized light energy is called as photons
- d. The energy of a packet of light depends on the frequency of light

$$E = h\nu$$

\_\_\_\_\_ now the quantum theory is known as photoelectric effect

- a. Light energy of photons is quantized in packets of energy depending on the frequency  $E = h\nu$
- b. Since photons are discrete packets of light
- c. As the wave number increases frequency of photons increases
- d. As frequency increases energy of photons increases
- e. As frequency increases energy of photons increases
- f. As frequency increases energy of photons increases

What happens for photons of light?

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What happens for photons of light?

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# Note Taking Guide 1302 Answers

**Sabine Zange**



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