

## 14

## THE BEHAVIOR OF GASES

## Chapter Test II

## A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct word or phrase in the space.

## Column I

1. The constant relating unit temperature change to the change in volume of a fixed amount of gas at constant pressure
2. The volume of a fixed amount of gas at constant pressure as the absolute temperature is changed
3. The ratio of the mass of a gas to its volume, independent of the actual amount of gas used
4. An experimental gas constant determined from the ideal gas law at low pressures
5. A constant of nature that has values of about 8.314 J/mol·K
6. The a parameter of gas constant equations for determining the gas molecular weight of a gas
7. An experimental gas constant derived from the ideal gas law at low pressures
8.  $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. The amount of heat enough to raise 1 g of water 1°C
10. A unit of energy, the joule

## Column II

- a. standard gas law
- b. ideal gas constant R
- c. density
- d. compressibility
- e. Boyle's law
- f. molar volume
- g. universal gas constant, pressure
- h. volume
- i. compressibility
- j. definition of joule

## B. Multiple Choice

Choose the correct answer and write its letter in the space.

1. According to Graham's law, the rate of effusion of a gas is proportional to the square root of its density. The rate of effusion of hydrogen is 9.0 times that of oxygen. The molecular weight of oxygen is 32. The molecular weight of hydrogen is:  
a. 16  
b. 4  
c. 2  
d. 1

# Prentice Hall Chemistry Chapter 13 And 14 Test Answers

**Scott C. Dulebohn**



**Prentice Hall Chemistry Chapter 13 And 14 Test Answers:**

## Decoding **Prentice Hall Chemistry Chapter 13 And 14 Test Answers**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Prentice Hall Chemistry Chapter 13 And 14 Test Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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