

Answer following



- 1. Substrate and temperature: 20°C 30°C 40°C
- 2. A graph of initial velocity: $\text{mg ml}^{-1} \text{min}^{-1}$ vs. mg ml^{-1} substrate concentration
- 3. A graph of initial velocity: $\text{mg ml}^{-1} \text{min}^{-1}$ vs. mg ml^{-1} substrate concentration

Calculate all values of K_m and V_{max} from all graphs in the diagram.
 (Note: K_m is the substrate concentration at which the reaction rate is half of V_{max} .)

- 1. $K_m = 10 \text{ mg ml}^{-1}$ $V_{max} = 10 \text{ mg ml}^{-1} \text{min}^{-1}$
- 2. $K_m = 10 \text{ mg ml}^{-1}$ $V_{max} = 10 \text{ mg ml}^{-1} \text{min}^{-1}$
- 3. $K_m = 10 \text{ mg ml}^{-1}$ $V_{max} = 10 \text{ mg ml}^{-1} \text{min}^{-1}$
- 4. $K_m = 10 \text{ mg ml}^{-1}$ $V_{max} = 10 \text{ mg ml}^{-1} \text{min}^{-1}$



Exercise 10

Graphs of initial velocity vs. substrate concentration are shown in the diagram. Calculate K_m and V_{max} from each graph.



Calculate K_m and V_{max} from each graph. (Note: K_m is the substrate concentration at which the reaction rate is half of V_{max} .)



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Guillaume Favre

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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