

Oxidation Number Exercise

Do not hand in this work sheet. When you are ready, you will be given an examination over this material. Complete the examination by yourself and hand it in to receive credit.

Purpose: This exercise is designed to teach the student how to assign oxidation numbers. Oxidation numbers are very important and are used for 1) naming compounds, 2) balancing oxidation-reduction reactions, 3) calculations in electrochemistry and other areas of chemistry.

Rule 0 The following rules are in the form of a hierarchy, that is, the first stated rule takes precedence over subsequent rules if a conflict arises.

Rule 1 The oxidation numbers for all the atoms in a neutral molecule must add up to 0. Similarly, the oxidation numbers for all the atoms of an ion must add up to the charge of the ion. (You are expected to recognize polyions. For the common polyions, know their charges and their names. A summary of the common polyions appears at the end of this exercise. The first step is, always, to determine what polyions are present.)

Rule 1a The oxidation number of elements in the elemental form is 0. (Note - this rule is a direct consequence of rule 1. How so?)

Rule 1b The oxidation number of any monatomic ion is the same as its charge. (See comment in 1a.)

Exercises - Give the oxidation number for the following atoms:

O₂ O = 0

F₂ F = 0

S₈ S = 0

Cl₂ Cl = 0

N₂ N = 0

Al Al = 0

Co²⁺ Co = +2

Mn²⁺ Mn = +2

Cl⁻ Cl = -1

Cr³⁺ Cr = +3

I₃⁻ I = -1/3

Hg₂²⁺ Hg = +1

Oxidation Number Exercise Answers

**Aadithi Dalvi, Dr. Chandrashekar
Bagawe**



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Physical Chemistry for JEE Advanced: Part 1, 3E (Free Sample) K. S. Verma, 2022-05-19 Physical Chemistry for JEE Advanced Part 1 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physical chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

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Chemistry Class XI - SBPD Publications Dr. Subhash Chandra Rastogi, ,Er. Meera Goyal, 2022-02-17 Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table

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Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table Basic Chemistry Concepts and Exercises John Kenkel, 2011-07-08 Chemistry can be a daunting subject for the uninitiated and all too often introductory textbooks do little to make students feel at ease with the complex subject matter Basic Chemistry Concepts and Exercises brings the wisdom of John Kenkel's more than 35 years of teaching experience to communicate the fundamentals of chemistry in a practical down to earth manner Using conversational language and logically assembled graphics the book concisely introduces each topic without overwhelming students with unnecessary detail Example problems and end of chapter questions emphasize repetition of concepts preparing students to become adept at the basics before they progress to an advanced general chemistry course Enhanced with visualization techniques such as the first chapter's mythical microscope the book clarifies challenging abstract ideas and stimulates curiosity into what can otherwise be an overwhelming topic Topics discussed in this reader friendly text include Properties and structure of matter Atoms molecules and compounds The Periodic Table Atomic weight formula weights and moles Gases and solutions Chemical equilibrium Acids bases and pH Organic chemicals The appendix contains answers to the homework exercises so students can check their work and receive instant feedback as to whether they have adequately grasped the concepts before moving on to the next section Designed to help students embrace chemistry not with trepidation but with confidence this solid preparatory text forms a firm foundation for more advanced chemistry training Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027 **NCERT Chemistry Class 11 - [CBSE Board]** Dr. S. C. Rastogi, ,Er. Meera Goyal, 2022-10-11 Syllabus Unit I Some Basic Concepts of Chemistry Unit II Structure of Atom Unit III Classification of Elements and Periodicity in Properties Unit IV Chemical Bonding and Molecular Structure Unit V States of Matter Gases and Liquids Unit VI Chemical Thermodynamics Unit VII Equilibrium Unit VIII Redox Reactions Unit IX Hydrogen Unit X s Block Elements Alkali and Alkaline earth metals Group 1 and Group 2 Elements Unit XI Some p Block Elements General Introduction to p Block Elements Unit XII Organic Chemistry Some Basic Principles and Techniques Unit XIII Hydrocarbons Classification of Hydrocarbons Unit XIV Environmental Chemistry Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table **Introduction to Chemistry** Charles H. Corwin, 1996 , Introduction to General, Organic, and Biochemistry Morris Hein, 1997 The authors recognize that both

science and mathematics may be daunting subjects for many students taking this course. With this in mind, they have anticipated where students might stumble and have paced and organized this text to help them through. Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career. The authors emphasize problem solving and provide a range of practice exercises. As in previous editions, the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry. In this edition, the first two sections have been revised primarily to improve explanations and include new pedagogical features. The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field. **NEET UG**

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nitrogen respectively Fourth section describes properties of various metals properties of acids and bases importance of nanotechnology in today s scenario Fifth section discusses about wider use of chemistry in agriculture food and medical sectors Sixth section is dedicated to polymers and its various varieties available in market Seventh section is related to chemical composition of fats and proteins The last eighth section detailed out thermodynamics and gas laws given by several scientists Further each section is divided in sub sections consisting detailed theory and practice questions The level of questions are easy to tough so that students may prepare not only for this exam but also other competitive exams such as UPSC CSAT State PSCs SSC JE etc The team OnlineVerdan have shown their best efforts to bring this unique book on e publication platform

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