

$$\Delta G^\circ = -2.86 \text{ kJ}$$

ह) मान निर्धारण E° की सहाय

$$\Delta G^\circ = -2.86 \text{ kJ} = -2.86 \times 10^3 \text{ J}$$

$$\log K_c = \frac{-\Delta G^\circ}{2.303RT}$$

$$= \frac{-(-2.86 \times 10^3)}{2.303 \times 8.314 \times 298} = 0.5074$$

अतः

$$K_c = 10^{0.5074}$$

$$K_c = 3.2$$

5. निम्नलिखित सेलों को 298 K पर सेल संकेतन एवं and

लिखिए



दिए गए हैं

$$E^\circ_{\text{Mg}^{2+}/\text{Mg}} = -2.37 \text{ V}$$

$$E^\circ_{\text{Co}^{2+}/\text{Co}} = -0.28 \text{ V}$$

$$E^\circ_{\text{Fe}^{2+}/\text{Fe}} = -0.44 \text{ V}$$

$$E^\circ_{\text{H}^+/\text{H}_2} = 0.00 \text{ V}$$

$$E^\circ_{\text{Sn}^{2+}/\text{Sn}} = -0.14 \text{ V}$$

$$E^\circ_{\text{Br}_2/\text{Br}^-} = 1.06 \text{ V}$$

$$E^\circ_{\text{Br}^-/\text{Br}_2} = -1.06 \text{ V}$$

$$E^\circ_{\text{H}_2/\text{H}^+} = 0.00 \text{ V}$$

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IM Harris



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