

Write Titrating Volume (Spreads: 1.00)

Name \_\_\_\_\_

Acetic Acid (pKa = 4.75) Titration with  $\text{NaOH}$

a.  $\text{CH}_3\text{COOH}$  volume  $25.00\text{ mL}$  conc.  $0.100\text{ M}$

b.  $\text{NaOH}$  volume  $25.00\text{ mL}$  conc.  $0.100\text{ M}$

c.  $\text{NaOH}$  conc.  $0.100\text{ M}$  compound that releases hydroxide  
 $\text{NaOH}$  conc.  $0.100\text{ M}$

Record Values of  $\text{pH}$  and  $\text{pH}$  for each and  $\text{pH}$  for each

$\text{pH}$   $\text{pH}$   $\text{pH}$   
 $\text{pH}$   $\text{pH}$   $\text{pH}$

Species	Add (M, Base (M, or Salt (M)	Acids with (M) (M or (M)	Conductivity (C or (C)
$\text{CH}_3\text{COOH}$	$\text{CH}_3\text{COOH}$	$\text{CH}_3\text{COOH}$	$\text{CH}_3\text{COOH}$
$\text{CH}_3\text{COO}^-$	$\text{CH}_3\text{COO}^-$	$\text{CH}_3\text{COO}^-$	$\text{CH}_3\text{COO}^-$
$\text{NaOH}$	$\text{NaOH}$	$\text{NaOH}$	$\text{NaOH}$
$\text{NaCH}_3\text{COO}$	$\text{NaCH}_3\text{COO}$	$\text{NaCH}_3\text{COO}$	$\text{NaCH}_3\text{COO}$

# Notes Taking Guide Episode 1101 Answers

**Annelies Wilder-Smith**



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web the pcl<sub>3</sub> lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh<sub>3</sub> chlorine cl

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