

Recap/Summary

1. Study of this _____ method _____ relationships in a _____ situation _____
2. Research on _____ behaviour _____ equated

2. $40g + 40g \rightarrow 2 \text{ MgO}$

2. Magnesium reacting with oxygen to produce magnesium oxide
 2. Magnesium reacting with oxygen to produce magnesium oxide

The _____ in a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

E.g. Problem

When aluminium reacts with oxygen to form aluminium oxide, the reaction is given by:



moles of Al = _____

$\frac{4}{4}$	$\frac{3}{3}$	$\frac{2}{2}$	$\frac{2}{2}$
$\frac{4}{4}$	$\frac{3}{3}$	$\frac{2}{2}$	$\frac{2}{2}$
$\frac{4}{4}$	$\frac{3}{3}$	$\frac{2}{2}$	$\frac{2}{2}$

2. If you start with 4 moles of Al, how many moles of Al_2O_3 will be produced?

2. 4 moles Al gives 2 moles Al_2O_3 and 3 moles O_2 is used to form 2 moles

Note Taking Guide Episode 801 Answers Key

Lei Huang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Huang'.

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