

Chemical Formula

Empirical relative number of atoms of each element in compound

Molecular : covalent bonding

Formula represents the number of atoms of each

element in a single molecule

Empirical : ionic bonding

Formula represents the simplest ratio of

positive and negative ions in a formula

unit

Electrolyte

An electrolyte that ions spontaneously ions

regardless

Electrolyte

1. Give away electrons

2. ions are positively charged

3. Cations

Electrolyte

1. Take electrons

2. ions are negatively charged

3. Anions

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