

Chemical Formula

Talking about proportions molecules and relative of atoms present in a compound

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of atoms of each
in a single molecule

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit

Atomic Mass

The average mass of atoms present in a sample
unit

Mass

m mass of an atom

M mass of a sample
unit

Amount

n amount of an atom

N amount of a sample

N_A amount

Note Taking Guide Episode 6chemical Formula Key

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Perry, Patricia A. Stockert, Amy Hall**

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