

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. CH_3COOH volume 25.00 mL and $[\text{CH}_3\text{COOH}] = 0.100\text{ M}$

b. CH_3COOH volume 12.50 mL and $[\text{CH}_3\text{COOH}] = 0.100\text{ M}$

c. CH_3COOH and CH_3COO^- composed for volume buffer
 12.50 mL and 12.50 mL and

Record Values of pH and pH for each and pH for each

$\text{pH} = 4.75$ $\text{pH} = 4.75$ $\text{pH} = 4.75$
 $\text{pH} = 4.75$ $\text{pH} = 4.75$ $\text{pH} = 4.75$

Species	Add NaOH , Base (pH or NaOH pH)	Acids with NaOH (pH or NaOH)	Conductivity (pH or NaOH)
CH_3COOH	pH	pH	pH
CH_3COO^-	pH	pH	pH
NaCl	pH	pH	pH
NaOH	pH	pH	pH

Note Taking Guide Episode 1101

E Durkheim



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