

7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa
7	hepta
8	octa
9	nona
10	deca

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation
anion

hydrates
anhydrates

electronegative
hydrogen ions

ionic
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate—**.
- When naming an ionic substance, the name of the **—cation—** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **—electronegative—** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **—hydrogen ions—**.
- The name of an acid is usually derived from the name of its characteristic **—anion—**.
- Prefixes are often used when naming **molecular** compounds.

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Number 1 2 Books and Pamphlets Including Serials and Contributions to Periodicals January December **Criminalistics**

James E. Girard,James Girard,2011-01-28 Criminal Investigations Forensic Science Study Guide/Selected Solutions

Manual Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book Environmental Chemistry Stanley E

Manahan,2017-02-24 With clear explanations real world examples and updated questions and answers the tenth edition of Environmental Chemistry emphasizes the concepts essential to the practice of environmental science technology and chemistry while introducing the newest innovations in the field The author follows the general format and organization popular in preceding editions including an approach based upon the five environmental spheres and the relationship of environmental chemistry to the key concepts of sustainability industrial ecology and green chemistry This readily adaptable text has been revamped to emphasize important topics such as the world water crisis It details global climate change to a greater degree than previous editions underlining the importance of abundant renewable energy in minimizing human influences on climate Environmental Chemistry is designed for a wide range of graduate and undergraduate courses in environmental chemistry environmental science and sustainability as well as serving as a general reference work for

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1. Understanding the eBook Prentice Hall Chemistry Test Prep Answers
 - The Rise of Digital Reading Prentice Hall Chemistry Test Prep Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Test Prep Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Test Prep Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Test Prep Answers
 - Personalized Recommendations
 - Prentice Hall Chemistry Test Prep Answers User Reviews and Ratings
 - Prentice Hall Chemistry Test Prep Answers and Bestseller Lists
5. Accessing Prentice Hall Chemistry Test Prep Answers Free and Paid eBooks
 - Prentice Hall Chemistry Test Prep Answers Public Domain eBooks
 - Prentice Hall Chemistry Test Prep Answers eBook Subscription Services
 - Prentice Hall Chemistry Test Prep Answers Budget-Friendly Options
6. Navigating Prentice Hall Chemistry Test Prep Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Prentice Hall Chemistry Test Prep Answers Compatibility with Devices
 - Prentice Hall Chemistry Test Prep Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Test Prep Answers
 - Highlighting and Note-Taking Prentice Hall Chemistry Test Prep Answers
 - Interactive Elements Prentice Hall Chemistry Test Prep Answers
 8. Staying Engaged with Prentice Hall Chemistry Test Prep Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Test Prep Answers
 9. Balancing eBooks and Physical Books Prentice Hall Chemistry Test Prep Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Test Prep Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Prentice Hall Chemistry Test Prep Answers
 - Setting Reading Goals Prentice Hall Chemistry Test Prep Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Prentice Hall Chemistry Test Prep Answers
 - Fact-Checking eBook Content of Prentice Hall Chemistry Test Prep Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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 - Interactive and Gamified eBooks

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