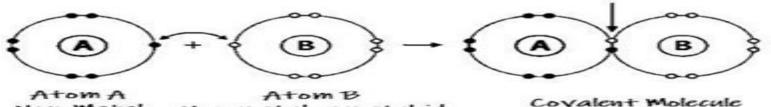
Covalent Bond

Unpaired valence electrons

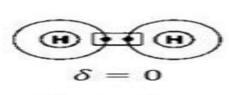
Non-Metal

Sharing of available valence electrons

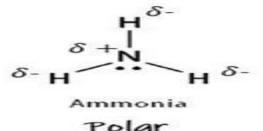


Electronegativity is either 0 or very less Noble Gas configuration

Non-metal or metaloid



Non Polar



Weak force of attraction

Ionic Bond > Covalent Bond

Bad Conductor of Electricity

Soluble in non-polar solvents and insoluble in polar solvents



Notes On Covalent Bond With Answers

PT Brinkman

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